



Molten Salt Spill Testing

Experiments to support accident progression analysis for MSR licensing

MSR Campaign Review Meeting

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Motivation and Objectives

Motivation

- Analysis of the effects of postulated accidents on safety is <u>required</u> to obtain NRC license for new nuclear reactors
- There is a lack of experimental data on processes that determine the consequences of potential molten salt reactor (MSR) accidents
 - Experimental data are needed by vendors preparing for the licensing process
 - Experimental data are needed by modelers to guide and advance model development

Objectives and Approach

- Provide technical bases for key processes included in mechanistic source term and accident progression models by conducting various laboratory-scale tests designed to highlight key processes
- Perform analyses using existing models to compare results with experimental data and to plan for large-scale integrated molten salt spill tests
 - e.g., MELTSPREAD to model molten salt spreading and heat transfer and perform sensitivity analysis on thermophysical properties

Molten Salt Spill Accident

• Common postulated accident scenario for many MSR concepts involves a rupture within the primary loop that leads to hot fuel salt spilling onto the primary containment floor



For MSR: salt spilling onto primary containment floor

Processes for which experimental data are needed to develop and validate models

- Spreading and flowing
 - On containment floor and through tubing into drain tank
- Heat transfer
 - Convection, conduction, radiation
- · Interactions with structural materials
 - Warping, corrosion
- Vaporization and condensation
- Aerosol formation
 - Due to splashing, spraying, bubble bursting, and vapor nucleation

Experimental Approach

Five modular laboratory test methods were designed to:

- Quantify individual processes important to accident progression modeling
- Provide data that can be used in individual process models
- Identify key factors and priorities for model development and future integrated tests

Test method descriptions:

- 1. Spreading and heat transfer of molten salt on sloped stainless steel sheets
- 2. Heat transfer of static molten salt pool contained in stainless steel
- 3. Aerosol generation due to splashing and vapor nucleation during a molten salt spill
- 4. Flowing and freezing of molten salt in stainless-steel tubing
- 5. Corrosion kinetics of stainless steel in molten salt

Salt compositions:

FY22

• NaCl-UCl₃ (eutectic composition)

FY21

- Pure FLiNaK
- FLiNaK doped with cesium and iodine as surrogate fission products (0.9 mol % CsF, 0.099 mol % CsI)

Test method 1: Molten salt spreading and heat transfer

- Significance of molten salt spreading and heat transfer to consequences of a molten salt spill accident:
 - Determines extent of radionuclide-bearing molten salt dispersal
 - Determines surface area of salt in contact with atmosphere (radionuclide vaporization)
 - Determines duration surface stays molten (radionuclide vaporization & substrate integrity)
 - Determines surface area of salt in contact with substrate (substrate integrity)
- Experiments were designed to quantify NaCI-UCI₃ spreading and provide temperature data to calculate heat loss from the salt

Test method 1: Molten salt spreading and heat transfer

Argon atmosphere glovebox Infrared & visible cameras

Schematic of test apparatus

Measurements

- Visible video of salt spreading
- Leading edge and covered area of salt on substrate vs. time (IR camera)
- Temperature of substrate underside, substrate surface, salt surface, and atmosphere
- Thickness of frozen salt collected after spreading





Variables tested

- Initial salt temperature (550 °C 750 °C)
- Pour mass

Quantifying molten NaCI-UCI₃ spreading for model validation

Initial salt temp: 750 °C, Mass salt poured: 65 g





- Results from IR video analysis provide data that can be directly compared with results from spreading and heat transfer models
 - · Leading edge vs. time
 - · Area vs. time
 - Salt surface temperature at each pixel vs. time



Modeling molten salt spreading and freezing using MELTSPREAD

- MELTSPREAD was developed at Argonne to model the one-dimensional flowing and freezing of molten corium
- MELTSPREAD was used to model molten salt spreading for the first time during FY21
- FY22 work is building upon the molten FLiNaK MELTSPREAD model developed during FY21 to incorporate
 processes identified in FLiNaK spreading experiments (i.e., crust formation at the salt/substrate interface) and to
 model a larger FLiNaK spill
 - Objective 1: to identify thermophysical properties that have the greatest effect on spreading behavior
 - Objective 2: to inform design of large scale salt spill tests

Spill conditions of model:

radial spreading of FLiNaK poured onto flat sheet



Thermophysical properties and initial conditions to vary for sensitivity analysis

• Viscosity

Density

- Pour mass
- Pour rate
- Surface tension Emissivity
- Thermal
- conductivity
- Heat capacity
- Heat of fusion

Test method 2: Heat transfer from static pool of molten salt

- Postulated accident scenario involves molten salt spilling onto the containment floor and forming a static pool
- Significance of heat transfer from static pool to consequences of molten salt spill accident:
 - Determines duration surface stays molten (radionuclide vaporization)
 - Determines substrate temperature (substrate integrity)
 - Determines duration molten salt is in contact with substrate (substrate corrosion)
- Experiments were designed to provide temperature data to calculate heat loss from the salt and the rate of conduction through the walls
 - Insulation minimizes cooling of substrate bottom
 - Natural convection cools the substrate walls
- Multiple variables can be tested systematically to determine their influence on the cooling behavior of the salt



Schematic of salt pool for model

Test method 2: Heat transfer from static pool of molten salt



Measurements

- Temperature of salt surface (IR camera)
- Temperature of salt (immersed thermocouple)
- Temperature of substrate underside (thermocouple)
- Temperature of substrate wall (IR camera)
- Thickness and mass of salt in beaker



Variables tested

- Beaker wall & bottom thickness (\mathcal{V}_4 ", \mathcal{V}_8 ", \mathcal{V}_{16} ")
- Beaker thermal mass (proportional to wall thickness)

Quantifying heat transfer from NaCI-UCI₃ for model validation

- Temperature of the salt surface was measured using an IR camera
- Tests with FLiNaK and NaCl-UCl₃ showed that salt near walls cools fastest and that FLiNaK is more efficient at retaining heat
- Spatially resolved salt surface temperature measurement can be used to create and validate heat transfer models for static pools of molten salt





Test method 3: Molten salt aerosol and splatter generation due to spilling

- Significance of aerosol and splatter formation to consequences of molten salt spill accident:
 - Determines dispersal of radionuclides (source term)
 - Radionuclide-bearing splatter
 - Radionuclide-bearing aerosols
 - Determines human health hazard if radionuclide-bearing aerosols are within respirable range
- Demonstrated technique during FY21 with FLiNaK
- Molten salt aerosol generation due to spilling NaCl-UCl₃ into a stainless steel catch pan is being tested in FY22

Measurements

- Visible video of molten salt splashing (slow motion)
- Splatter abundance, size distribution, and composition
- Temperature of atmosphere within spill containment box and at underside of catch pan
- Aerosol composition collected on filters (0.45 µm) by ICP-MS and collected on adhesive tape by SEM-EDS



Variables to test

- Initial salt temperature (550 °C, 650 °C, 750 °C)
- (Potentially) presence of CsCl and Csl surrogate fission products

Test method 3: Molten salt aerosol and splatter generation due to spilling

- Coupons with adhesive are effective at collecting splatter on the walls of the catch pan for analysis
 - Abundance and size distribution is determined by analyzing images taken by visible camera
 - Composition is determined by using SEM-EDS
- PTFE filter and Coupon 7 (below filter) are effective at collecting aerosol particles for analysis
 - Composition of aerosol particles on filter is determined by using ICP-MS and composition of aerosol particles on Coupon 7 is determined by using SEM-EDS
 - Size of individual aerosol particles on Coupon 7 is determined by using SEM-EDS



FLiNaK splatter collected on coupon



SEM image of cesium and iodine containing particles on Coupon 7



Summary of accomplishments and milestones

Summary of accomplishments

- Methods developed to quantify individual processes are being employed with a representative chloride salt (NaCI-UCI₃) to provide data for model development
 - Conducted molten salt spreading tests on a stainless steel sheet to quantify spreading behavior
 - Conducted molten salt heat transfer tests in a stainless steel beaker to provide temperature measurements with high spatial and temporal resolution
 - Tests on splatter and aerosol generation are in progress
- MELTSPREAD model of the radial spreading of FLiNaK on stainless steel is being developed to:
 - Incorporate crust formation at the salt/substrate interface
 - Perform a sensitivity analysis of thermophysical properties and initial conditions for a large scale FLiNaK spill

Milestone Number	Title	Due
M4RD-22AN0602092	MELTSPREAD modeling report	6/24/22
M3RD-22AN0602091	Report and data package of results from spill tests	9/23/22

Upcoming milestones – on schedule

Future Work

Future work (near term)

- Conduct individual process tests at laboratory scale, including those that emphasize:
 - Molten salt aerosol studies
 - Complex salt compositions (i.e., those containing surrogate fission products) and environmental conditions (i.e., presence of humidity and oxygen)
 - Irradiated salts
- Integrate ongoing MSR campaign work (e.g., sensors for off gas components) in experiments to understand consequences of MSR salt spill accidents
- Design large scale integrated tests with input from MSR campaign participants, modelers, and MSR developers

Future work (long term)

• Construct large scale integrated salt spill test facility and conduct tests using representative fluoride and chloride salts





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