



Update on Molten Salt Database— Thermophysical (MSD-TP)

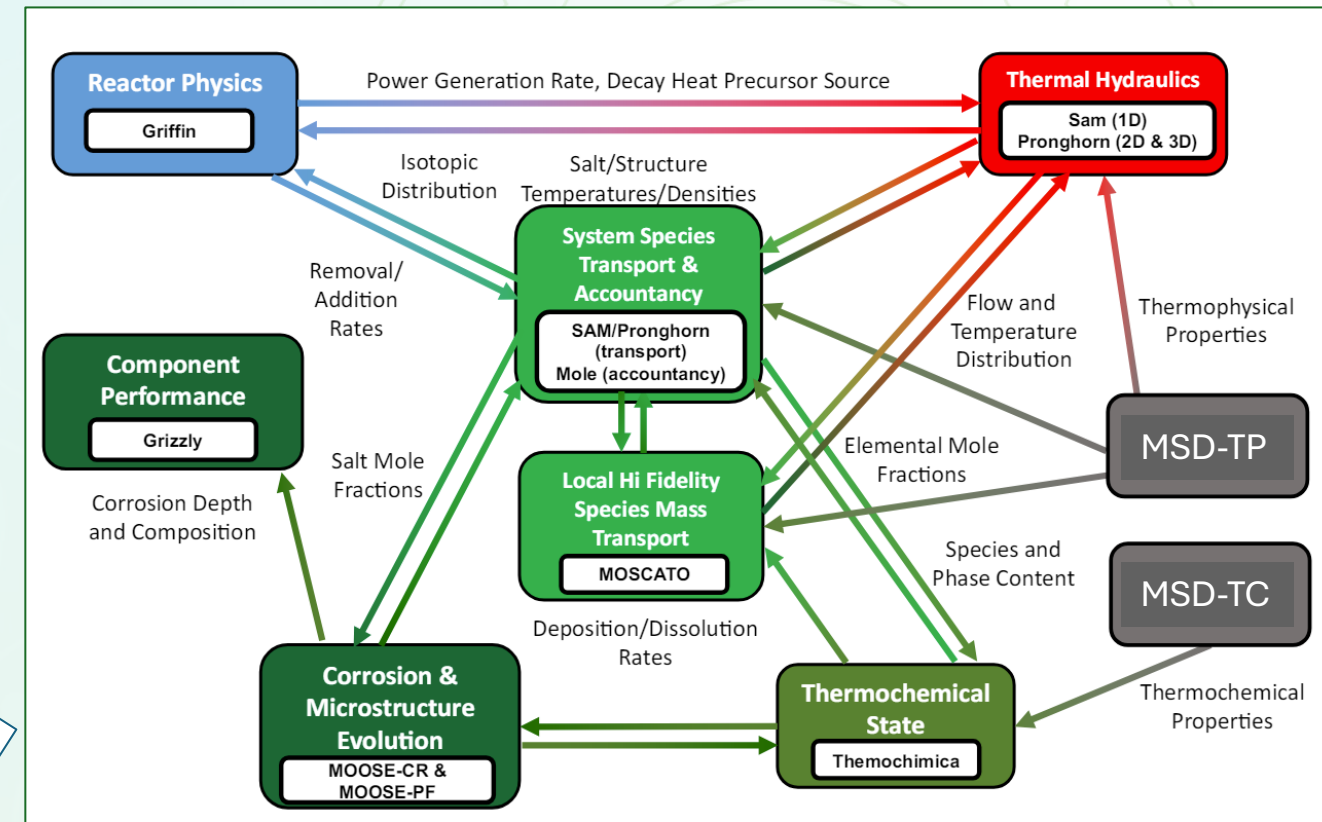
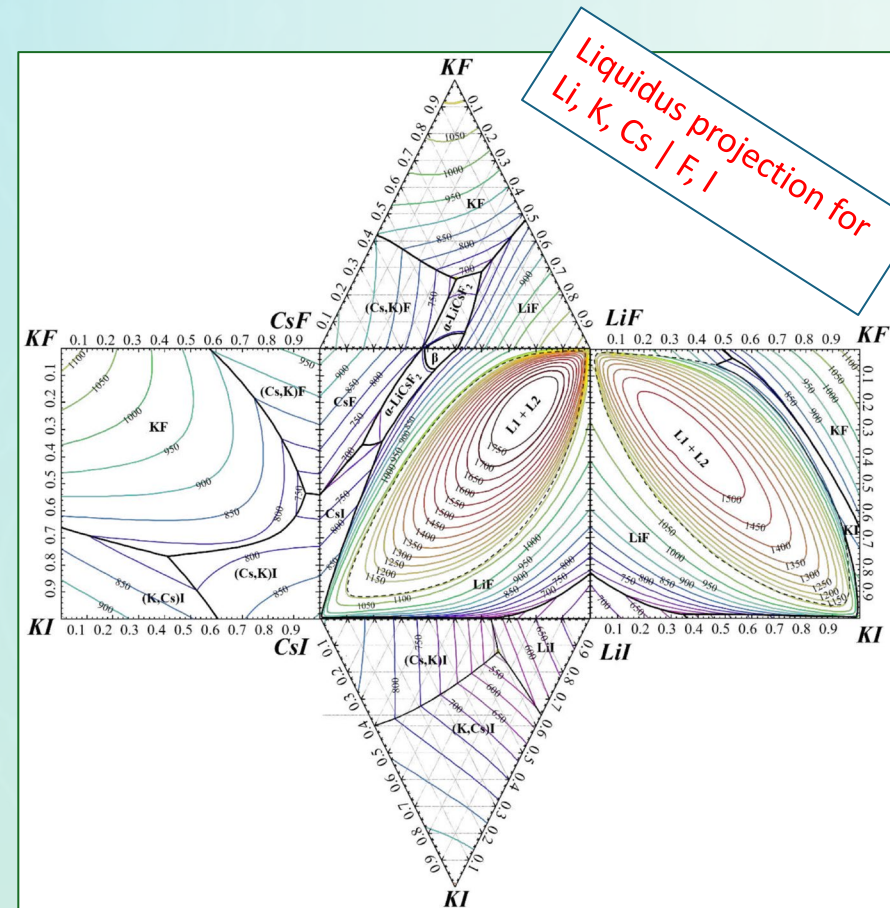
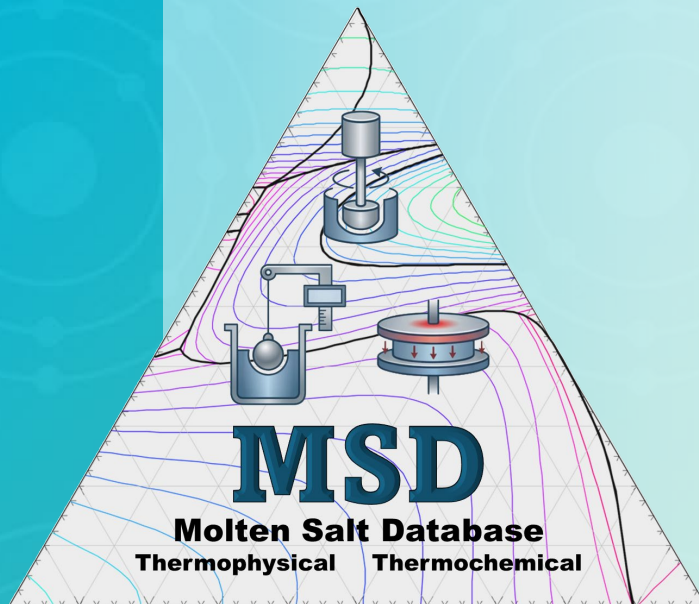
Anthony Birri, Ryan Chesser, Shane Henderson, Jacob Numbers, Difan Zhang,
Kyoung Lee, and Vanda Glezakou

Molten Salt Reactor Program – Annual Review

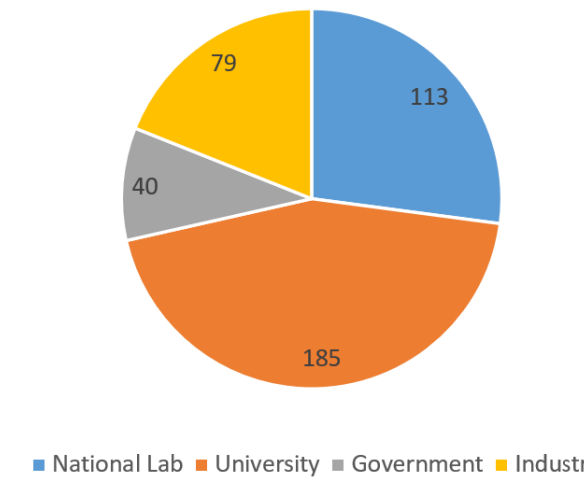
April 21

Overview of MSD

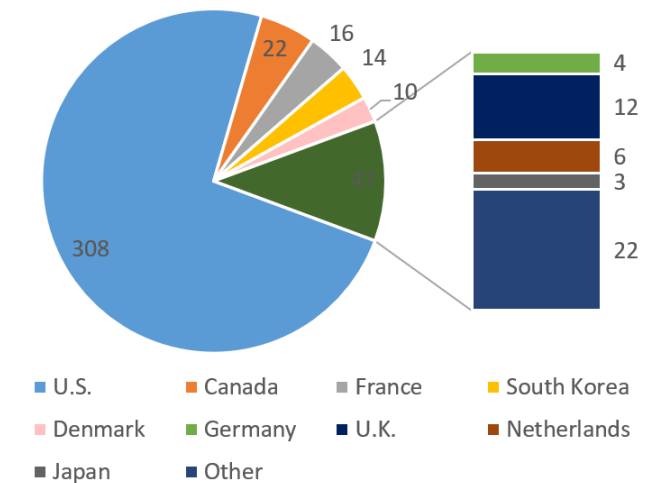
- The Molten Salt Database–Thermochemical (MSD-TC) and Molten Salt Database–Thermophysical (MSD-TP) are available via the ORNL/ITSD Gitlab Server.



MSD Subscribers by Institution Type



MSD Subscribers by Country

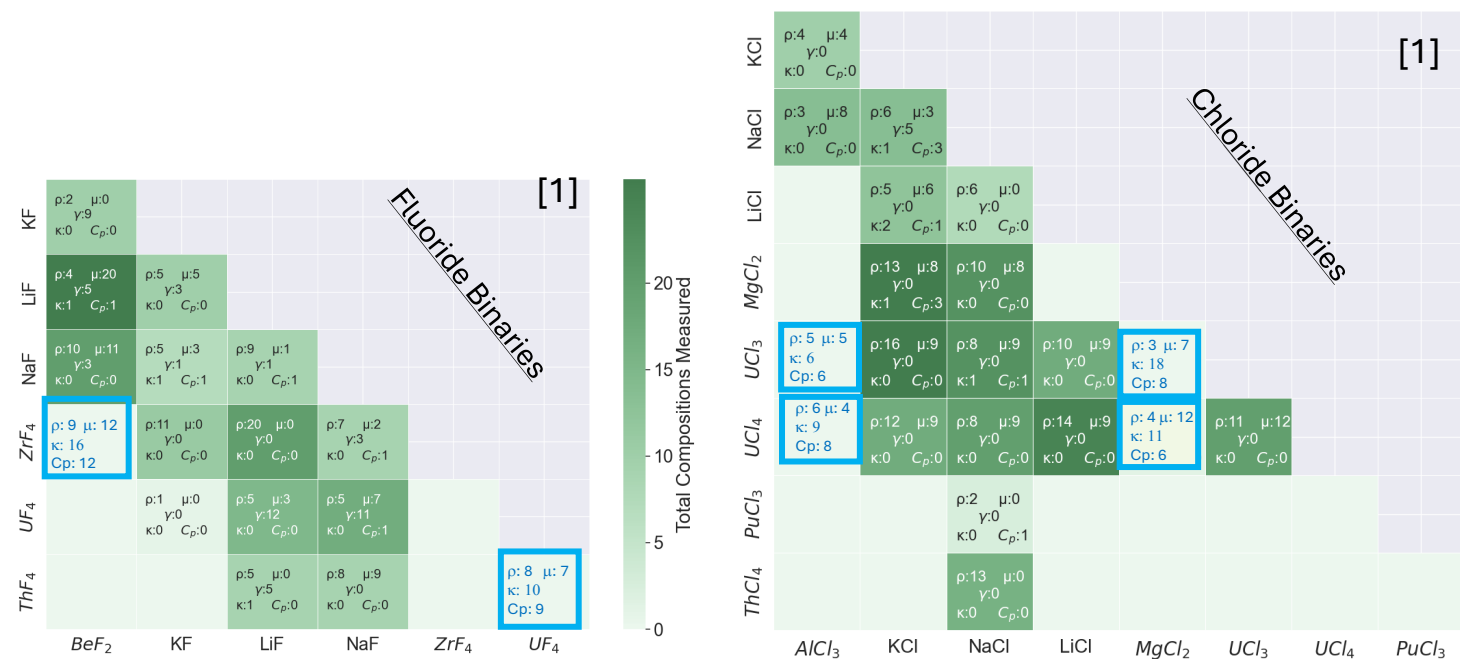


- [1] INL/RPT-22-70694
- [2] ORNL/TM-2025/4159
- [3] Dixon, Clara M., et al. The Journal of Chemical Thermodynamics (2025): 107548.



MSD-TP 4.0 Overview

- As per the current version release (4.0) There are **976 entries**
 - Newest version contains surface tension data
 - MSD-TP is driven by a roadmap laid out in 2020: [ORNL/SPR-2020/1865](#)**
 - In blue: literature entries using LLM's (Zhang, Glezakou)**



Version:	1.0	2.0	2.1	3.0	3.1	4.0
Release Date:	Feb-21	Mar-22	Dec-22	Mar-24	Dec-24	Oct-25
Number of Entries:	62	273	448	799	820	976
RK Binary System Characterizations:	0	0	21	29	47	47
Saline Integration of RK Density:	No	No	Yes	Yes	Yes	Yes
Saline Integration of RK Viscosity:	No	No	No	No	Yes	Yes
Number of Properties Considered:	4	4	4	4	4	5

Pure Compounds

Salt	ρ	μ	κ	c_p	γ
AlCl3	1	1	0	1	0
BeCl2	1	0	0	0	0
BeF2	1	1	1	1	0
CaCl2	1	1	1	1	1
CaF2	1	1	1	1	1
GdCl3	1	1	0	0	1
GdF3	0	0	0	0	0
KCl	1	1	1	1	1
KF	1	1	1	1	1
LaCl3	1	1	0	0	1
LaF3	1	0	0	1	0
LiCl	1	1	1	1	1
LiF	1	1	1	1	1
MgCl2	1	1	1	1	1
MgF2	1	1	1	0	1
NaCl	1	1	1	1	1
NaF	1	1	1	1	1
NdCl3	1	1	0	0	1
NdF3	0	0	0	1	0
NpCl3	0	0	0	0	0
NpF3	0	0	0	0	0
PuCl3	0	0	0	1	0
PuF3	0	0	0	1	0
SrCl2	1	1	1	0	1
SrF2	1	1	1	0	1
ThCl4	1	0	0	0	0
ThF4	1	0	0	0	1
UCl3	1	1	0	1	1
UCl4	1	1	0	0	1
UF3	0	0	0	1	0
UF4	1	1	0	1	1
ZrCl4	1	1	0	0	0
ZrF4	1	0	0	0	1
RbCl	0	0	0	0	1
CsCl	0	0	0	0	1
BaCl2	0	0	0	0	1
PrCl3	0	0	0	0	1
RbF	0	0	0	0	1
CsF	0	0	0	0	1
BaF2	0	0	0	0	1
UF6	0	0	0	0	1

Ternaries

Salt	ρ	μ	κ	c_p	γ
AlCl3-LiCl-NaCl	10	10	0	0	0
BeF2-KF-NaF	1	1	0	1	0
BeF2-LiF-NaF	5	4	0	5	0
BeF2-LiF-ThF4	3	2	0	0	44
BeF2-LiF-UF4	36	38	2	0	1
BeF2-LiF-ZrF4	1	0	0	0	0
BeF2-NaF-UF4	78	70	0	0	1
KCl-LiCl-NaCl	4	0	0	0	0
KCl-LiCl-UCl3	18	0	0	0	0
KCl-LiCl-UCl4	18	0	0	0	0
KCl-MgCl2-NaCl	1	1	1	1	0
KCl-NaCl-UCl3	18	20	0	1	0
KCl-UCl3-UCl4	32	0	0	0	0
KF-LiF-NaF	1	1	1	1	1
KF-MgF2-NaF	1	0	1	1	0
KF-NaF-UF4	4	2	3	3	0
KF-NaF-ZrF4	1	1	0	1	0
NaCl-PuCl3-UCl3	3	0	0	0	0
MgCl2-NaCl-PuCl3	2	0	0	0	0
LiCl-NaCl-UCl3	18	0	0	0	0
LiCl-NaCl-UCl4	18	0	0	0	0
LiCl-UCl3-UCl4	21	0	0	0	0
LiF-NaF-ZrF4	10	1	1	2	11
LiF-ThF4-UF4	1	0	0	0	0
NaCl-UCl3-UCl4	21	40	0	0	0
NaF-UF4-ZrF4	5	4	0	6	0
RbF-UF4-ZrF4	2	2	0	2	0
CaCl2-LaCl3-NaCl	0	0	0	0	1
CaF2-LiF-NaF	0	0	0	0	1
KF-LiF-ZrF4	0	0	0	0	1

Quaternaries

Salt	ρ	μ	κ	c_p	γ
BeF2-LiF-NaF-UF4	1	1	0	1	0
BeF2-LiF-ThF4-UF4	1	1	1	1	32
BeF2-LiF-UF4-ZrF4	1	0	1	0	0
BeF2-NaF-UF4-ZrF4	1	0	0	0	0
KF-LiF-NaF-UF4	2	2	0	2	0
KF-NaF-UF4-ZrF4	0	0	0	2	0
LiF-NaF-UF4-ZrF4	1	1	0	2	0
BeF2-NaF-ThF4-UF4	0	0	0	0	1

[1] ORNL/TM-2025/4159



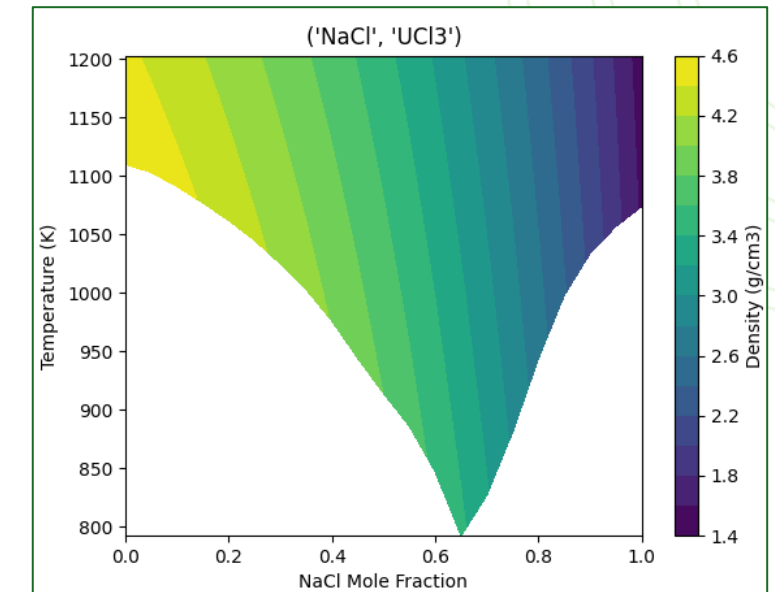
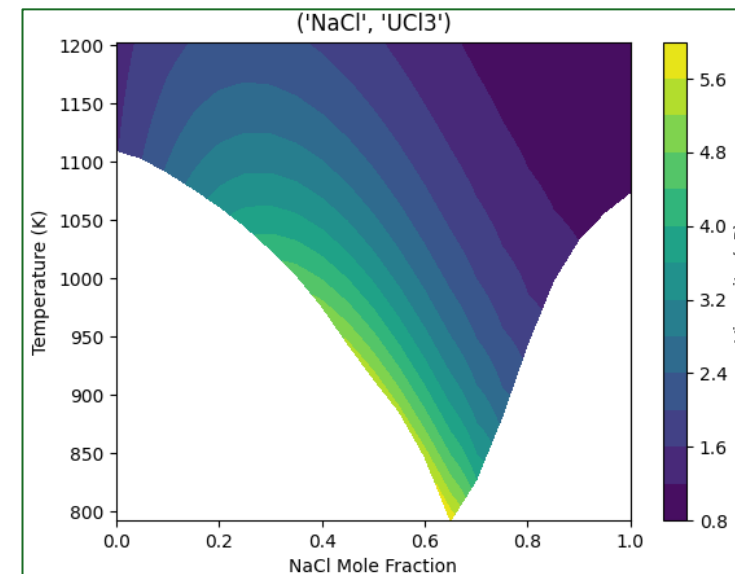
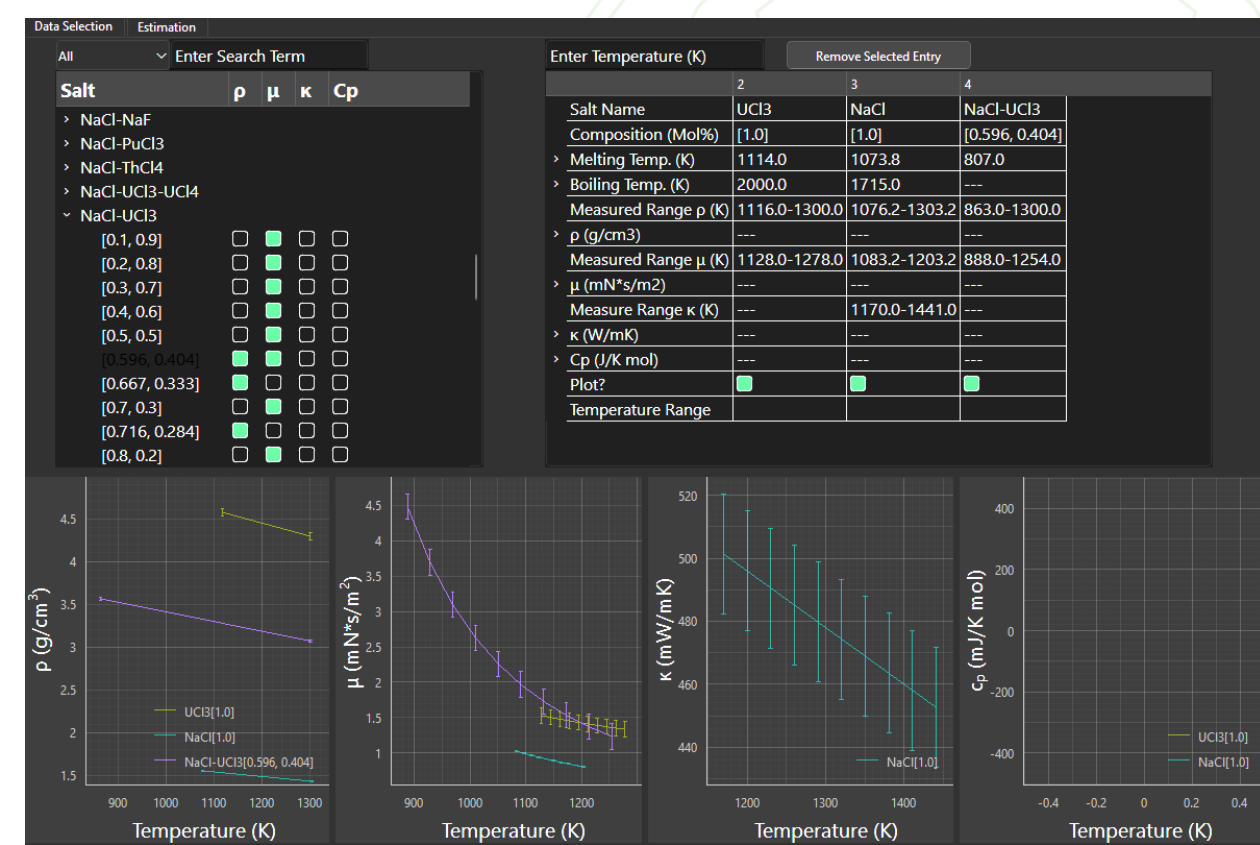
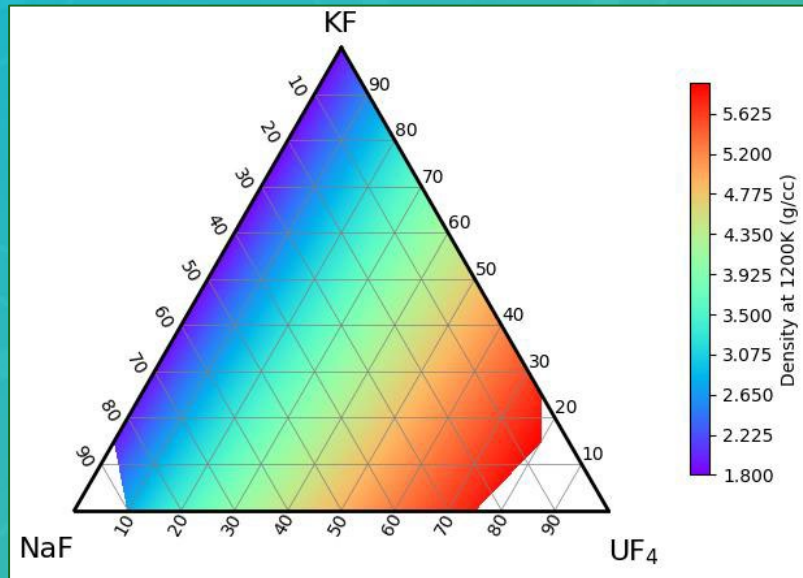
Molten Salt Reactor
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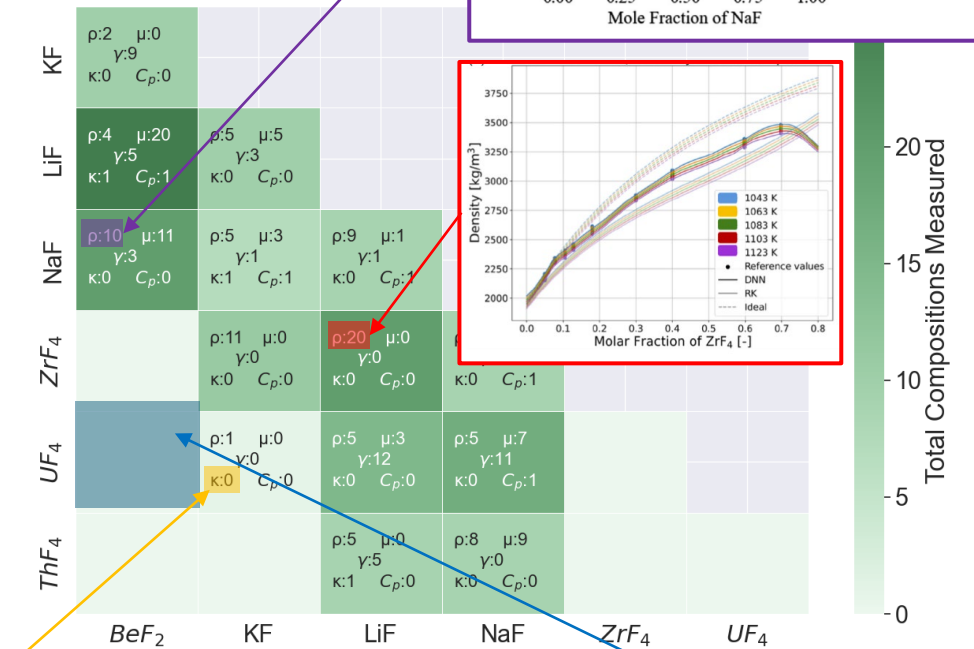
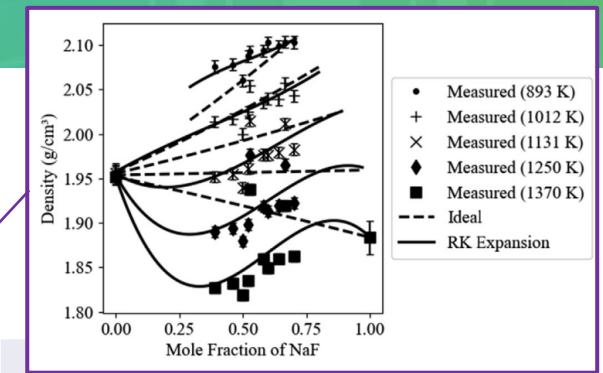
Interfaces: Saline, GUI

- Saline is a stable c++ API with a python wrapper available. Implemented in MOOSE as an object.
 - Ideal for code integration, accessing data rapidly
- GUI can be downloaded as a .exe file
 - Predictive models are outputted over ranges deemed valid via thermochemica-generated look-up table

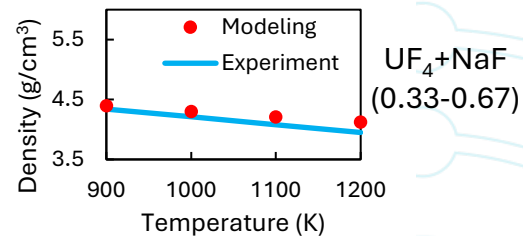


Bridging the gaps; a multifaceted solution

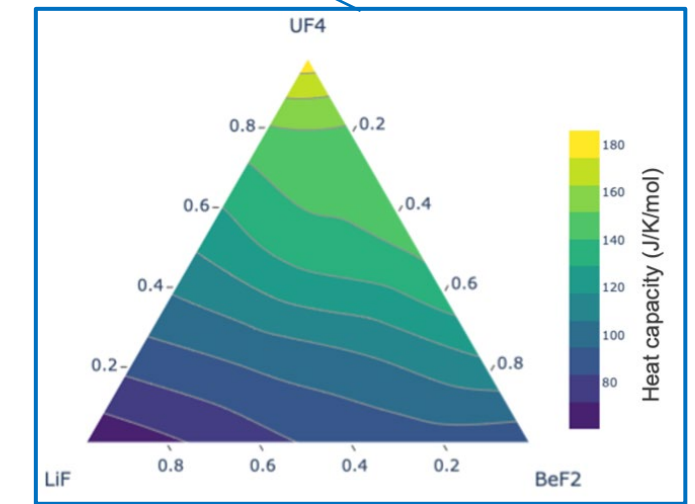
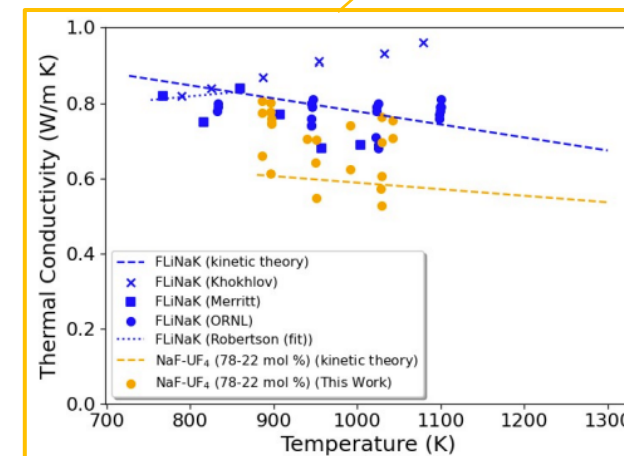
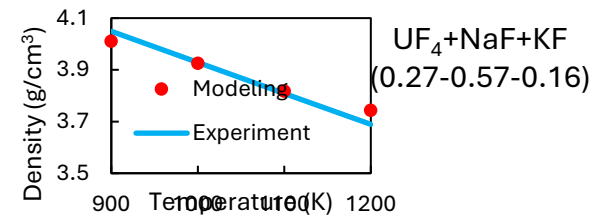
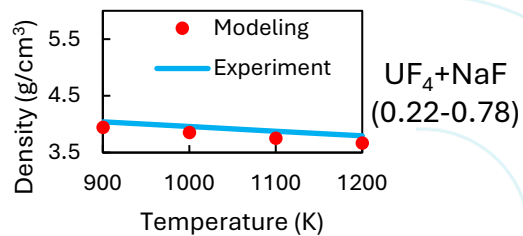
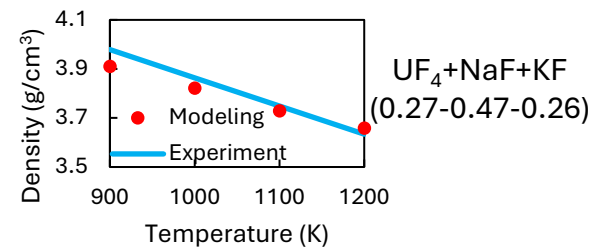
- Several gaps to fill in the database; how do we address this?
- Multifaceted approach to solving this going forward:
 - Traditional solution models (such as RK) to interpolate and extrapolate [1]
 - Transfer Learning with ML [2]
 - Direct calculation with MD [3,4]
 - Experimental and theoretical collaboration through the MSR program [5]
 - Other theoretical approaches (such as kinetic theory)



UF₄+NaF [4]



UF₄+NaF+KF [4]



- [1] Birri, Anthony, et al. Chemical Engineering Science 260 (2022): 117954.
 [2] Barra, Julian, et al. ChemPhysChem 26.23 (2025): e202500273.
 [3] Wang et al. "Ab-initio molecular dynamics simulations of LiF-BeF₂-UF₄ molten salts". LANL.
 [4] Zhang, Glezakou MD properties and transport in NaF-UF₄, NaF-KF-UF₄, in preparation
 [5] ORNL/TM-2025/4057

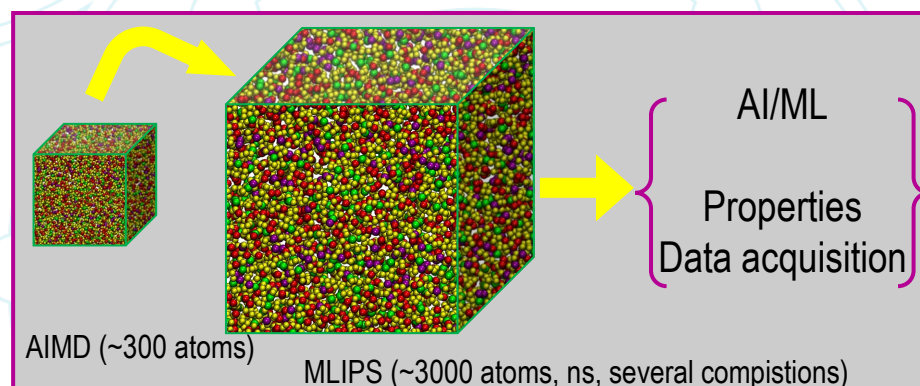


A note on uncertainty

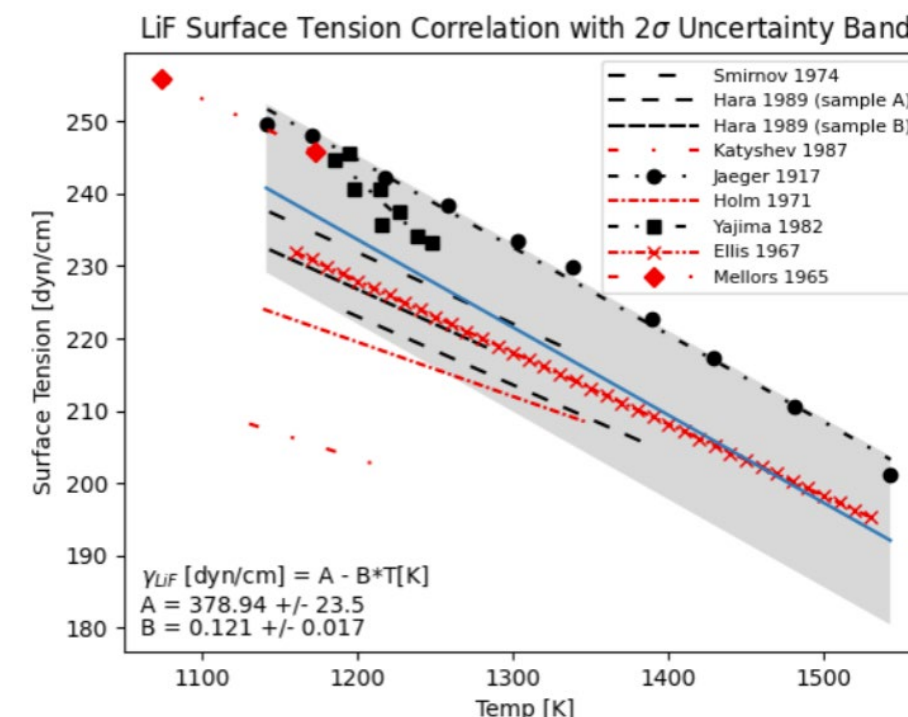
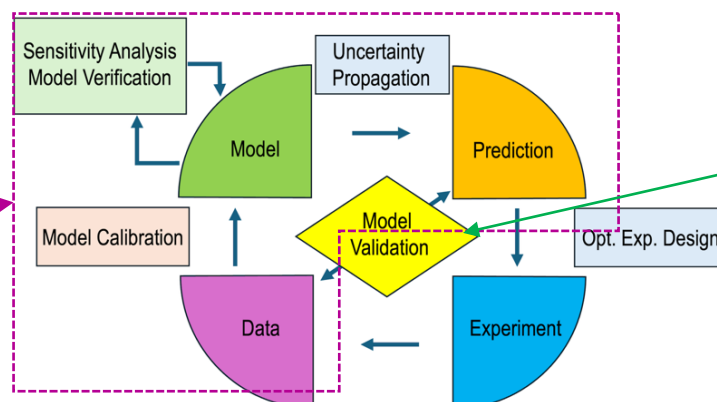
Quality Ranking of Unary Fluoride Salt Property Data in MSTDB-TP

- MSD-TP currently contains experimental data, with the plan to integrate MD-generated data in the next iteration
- For experimental uncertainties, sometimes the uncertainties are well characterized
- However, sometimes data is provided without error bars; then we rely on conservative assumptions
- As we integrate MD data, we are working with experts in NEAMS to establish a strategy
 - Data quality assessments will help in this effort
- General observation: Uncertainty quantification relies on statistical methods, hence large, well-curated datasets are required [2]
 - Computed data with well-validated models can fill these gaps

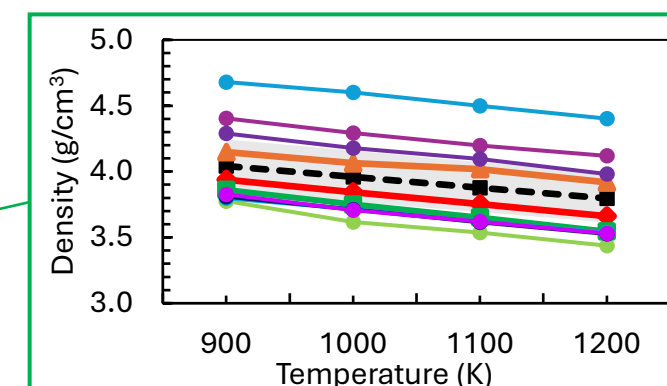
[2] Zhang, Rousseau, Glezakou ACS Catal., 14(10), pp.8073-8086



UQ Workflow for Models and Experiment (after Ralph Smith, NCSU)



[1] Chesser et al. "Molten Salt Surface Tension - Methods and Correlations". Under review



Thermal Conductivity Modeling

- The lack of experimental thermal conductivity data necessitates a more theoretical model
- We have shown that a kinetic theory provide accurate data for actinides
 - Existing frameworks underpredict thermal conductivity
- This modeling approach relies on characterizing phonon transport
 - In some cases, where MD data is sufficient, phonon transport can be characterized by PDFs

Kinetic Theory Model (purely theoretical)

$$\kappa = K(1 - \delta)k_B \left(\frac{N_A n}{V(T_m)} \right) c_0(T_m) \left(1 - \alpha(T_m) \left(\gamma(T_m) + \frac{1}{3} \right) (T - T_m) \right)$$

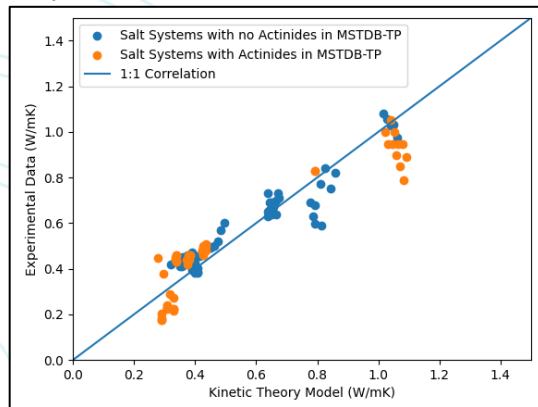
Scaling factor for ionic mass differences

Boltzmann constant

Phonon mean free path

Group velocity

Temperature dependence (relies on heat capacity)

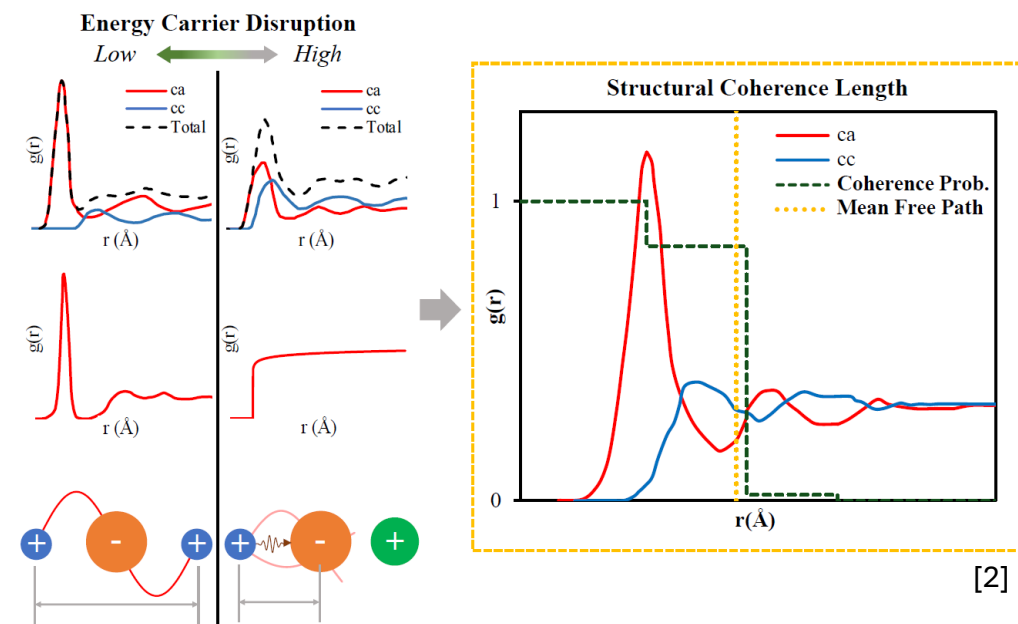


[1]

[1] ORNL/TM-2025/4159

[2] Numbers et al. "Structural Coherence Model for Predicting Molten Salt Thermal Conductivity Informed by the Pair Distribution Function." Under Review.

Structural Coherence Model (tied to PDFs)



[2]



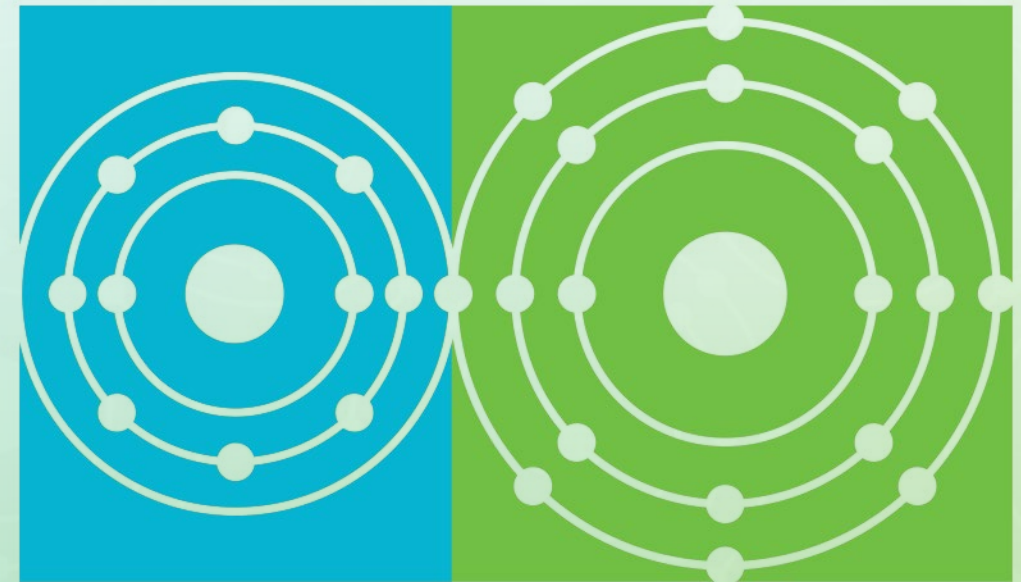
Acknowledgements

- Ted Besmann, Juliano Shorne-Pinto, and Jorge Paz Soldan Palma from USC for fruitful collaborations on MSD
- Troy Munro and Jacob Numbers at BYU for thermal conductivity modeling collaborations
- Chao Jiang and Gaoxue Wang at INL and LANL for insightful discussions regarding MD integration

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Questions?

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