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Overview of MSTDB-TP

Anthony Birri, Nicholas Termini, Shane Henderson,

N. Dianne Bull Ezell

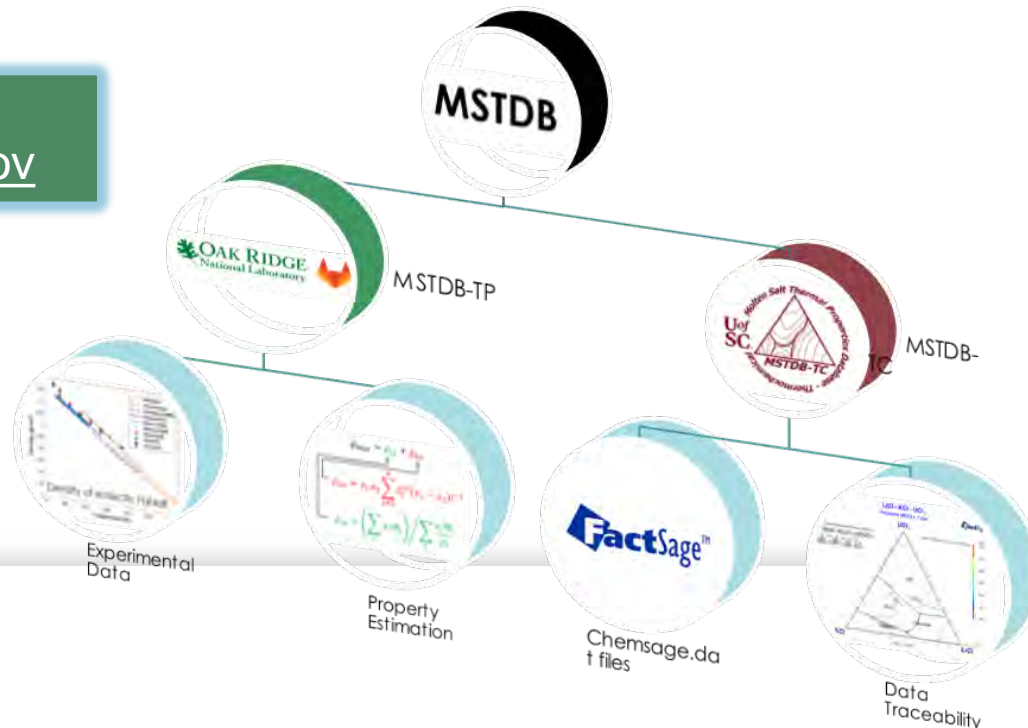
Oak Ridge National Laboratory

Annual MSR Campaign Review Meeting 2-4 May 2023

How to get Access

- The databases and associated documents are hosted on a publicly accessible, permission-protected server at Oak Ridge National Laboratory (ORNL):
- <https://code.ornl.gov/neams/mstdb/>
- Access requires an ORNL XCAMS account and an MSTDB membership, which once granted will allow downloading of all files.

Contact:
mstdb@ornl.gov



Molten Salt Thermal Properties Databases

The Molten Salt Thermal Properties Database–Thermochemical (*MSTDB-TC*) and Molten Salt Thermal Properties Database–Thermophysical (*MSTDB-TP*) databases are now available for public use. *MSTDB-TC* contains Gibbs energy models and values for molten salt components and related systems of interest with respect to molten salt reactor technology. *MSTDB-TP* consists of tabulated thermophysical properties and relations for computing properties as a function of temperature or composition.

MSTDB-TC thermodynamic information resides in files in the “Chemsage” .dat (ASC II) format for use with the FactSage® commercial package of thermodynamic codes and compatible with the open-source equilibrium code Thermochemica.

- Separate files are provided for chloride- and for fluoride-based systems.
- Changes by FactSage® developers have resulted in the need to provide files readable by FactSage® Ver. 8.0 or lower and FactSage® Ver. 8.1 or higher, as noted in the documentation that will accompany the database download.
- Additional files include those for tracing all data sources and a library of published phase diagrams together with companion *MSTDB-TC*-computed phase diagrams to allow the user to compare model results with those previously reported.

MSTDB-TP contains sets of referenced values and relations for thermophysical properties including density, thermal conductivity, viscosity, heat capacity, and related optical properties. The database is maintained as a csv file of the salt systems thermophysical property data, uncertainty (when available), and literature references.

Accessing MSTDB: The databases and associated documents are hosted on a publicly accessible, permission-protected server at Oak Ridge National Laboratory (ORNL): <https://code.ornl.gov/neams/mstdb/>. Access requires an ORNL XCAMS account and an MSTDB membership, which once granted will allow downloading of all files.

- XCAMS account creation
 - Go to <https://xcams.ornl.gov>
 - Select “I need an account.”
 - Read and acknowledge the User Agreement
 - Enter your email address and username following the guidelines on the page.
 - Enter “Personal Information” and “Contact Information” per the guidelines
 - Create an XCAMS password according to the guidelines provided on the page.
 - On the final step, note the activation sequence box at mid-page. Wait until each action item turns green and the box heading reads “Transactions Complete”
 - Log into <https://code.ornl.gov> using your new XCAMS username and password
- Request MSTDB membership
 - Send an email to mstdb@ornl.gov with “MSTDB Access Request” as subject
 - Include your XCAMS ID and brief summary of the purpose for your request

MSTDB-TC is a copyrighted database available license-free, and cannot be sold all or in part. Development is supported by the U.S. Department of Energy, Office of Nuclear Energy Molten Salt Reactor Campaign, Nuclear Energy Advanced Modeling and Simulation Program, and Nuclear Energy University Programs. *MSTDB-TP* is a product of Department of Energy MSR-related research programs.

Contacts: *MSTDB-TC* Ted Besmann (besmann@sc.edu)
MSTDB-TP Dianne Ezell (bullnd@ornl.gov)

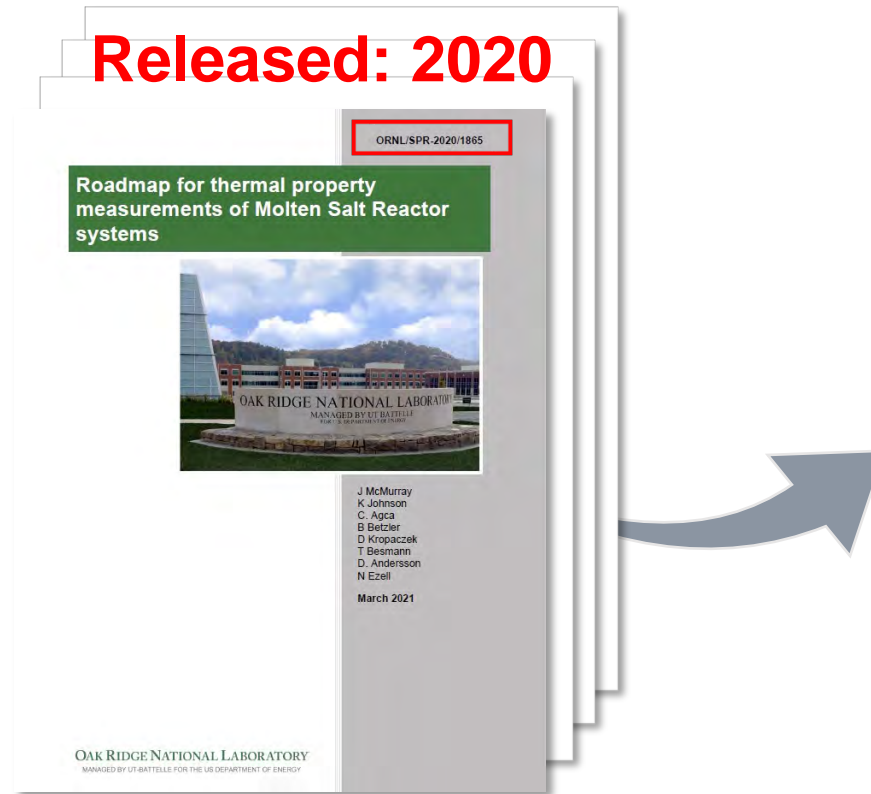
MSTDB-TP Overview

Please submit your publications with measurement data (including uncertainty)

- **The Molten Salt Thermal Property Database– Thermophysical (MSTDB-TP) contains empirical relations for the following properties:**
 - Melting and boiling points
 - Density
 - Viscosity
 - Heat Capacity
 - Thermal Conductivity
- **As per the current version release (v2.1) There are 448 entries, including:**
 - 33 pure compounds
 - 243 pseudo-binaries
 - 166 pseudo-ternaries
 - 6 pseudo-quaternaries
- **Each property entry in the database includes a margin of experimental error**
 - Determined on a case-by-case basis
 - This list is constantly expanding. The data is based on the outputs of 140+ independent experimental studies in literature.

MSTDB-TP Expansion Efforts

- Outdated figures should no longer be referenced
- Still flowing roadmap guidance!



McMurray, Jake W., et al. Roadmap for thermal property measurements of Molten Salt Reactor systems. No. ORNL/SPR-2020/1865. Oak Ridge National Lab.(ORNL), Oak Ridge, TN (United States), 2021.

Key
A: Density
B: Viscosity
C: Thermal conductivity
D: Heat capacity

KF	B						
LiF	A, B, D	A, B, C, D					
NaF	A, B	A, B, C	A, B, C, D				
ThF4		B	A, B, D	A, B			
UF3							
UF4		A, B	A, B	A, B			
ZrF4			A	A, B, D			
BeF2	KF	LiF	NaF	ThF4	UF3	UF4	
KCl	A, B						
LiCl	A	A, B, D					
MgCl2		A, B, D	A, B				
NaCl	A, B	A, B, D	A, D	A, B			
PuCl3							
ThCl4		A		A			
UCl3		A, B	A	A, B			
UCl4		A, B	A	A, B			
ZrCl4		A					
AlCl3	KCl	LiCl	MgCl2	NaCl	PuCl3	ThCl4	UCl3

MSTDB-TP Expansion Efforts

Available @ mstdb.ornl.gov

- MSTDB-TP has undergone 2 major expansion efforts:**

- 1.0 to 2.0 (68 entries to 273 entries)
- 2.0 to 2.1 (273 entries to 448 entries)

- These expansions incorporate replacements of old datasets as well**

- E.g. recent literature has suggested UCl3 and relevant mixtures has a lower thermal expansion coefficient than previously understood

- MSTDB-TP is being expanded for later releases**

- This includes new pseudo-binary and higher order system data that exist in literature and need evaluated
- MSTDB-TP will also include new data of new systems as it is published

- MSTDB-TP is intending on including surface tension data in the future**

- There is a significant body of literature already evaluated and tabulated

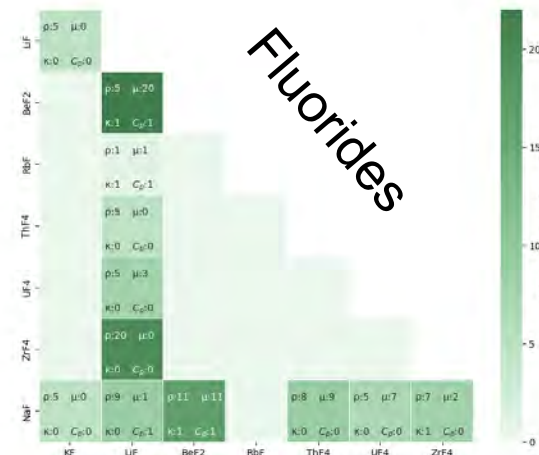
Pure:

Salt	Measurements			
	ρ	μ	κ	c_p
AlCl3	1	1	0	1
BeCl2	1	0	0	0
BeF2	1	1	1	1
CaCl2	1	1	1	1
CaF2	1	1	1	1
GdCl3	1	1	0	0
GdF3	0	0	0	0
KCl	1	1	1	1
KF	1	1	1	1
LaCl3	1	1	0	0
LaF3	1	0	0	1
LiCl	1	1	1	1
LiF	1	1	1	1
MgCl2	1	1	1	1
MgF2	1	1	1	0
NaCl	1	1	1	1
NaF	1	1	1	1
NdCl3	1	1	0	0
NdF3	0	0	0	1
NpCl3	0	0	0	0
NpF3	0	0	0	0
PuCl3	0	0	0	1
PuF3	0	0	0	1
SrCl2	1	1	1	0
SrF2	1	1	1	0
ThCl4	1	0	0	0
ThF4	1	0	0	0
UCl3	1	0	0	1
UCl4	1	0	0	0
UF3	0	0	0	1
UF4	1	1	0	1
ZrCl4	1	1	0	0
ZrF4	1	0	0	0

Binary:



Living database



Ternary:

Salt	Measurements			
	ρ	μ	κ	c_p
KCl-LiCl-NaCl	4	0	0	0
LiCl-NaCl-AlCl3	10	10	0	0
LiF-BeF2-ThF4	3	2	0	0
LiF-BeF2-ZrF4	1	0	0	0
LiF-NaF-BeF2	1	1	0	0
LiF-NaF-KF	1	1	1	1
LiF-BeF2-UF4	36	36	0	0
NaF-BeF2-UF4	79	71	0	0
NaF-KF-BeF2	1	1	0	0
NaF-KF-MgCl2	1	0	0	0
NaF-KF-UF4	1	1	1	1
NaF-KF-ZrF4	1	1	0	0
NaF-LiF-BeF2	4	4	0	0
NaF-LiF-ZrF4	10	1	0	1
NaF-ZrF4-UF4	5	3	2	3
RbF-ZrF4-UF4	2	2	1	1

Quaternary:

Salt	Measurements			
	ρ	μ	κ	c_p
LiF-BeF2-UF4-ThF4	1	1	0	0
LiF-BeF2-ZrF4-UF4	1	0	0	0
NaF-LiF-BeF2-UF4	1	1	0	0
NaF-LiF-KF-UF4	2	2	1	1
NaF-LiF-ZrF4-UF4	1	1	0	1

MSTDB-TP Expansion Efforts

Available @ mstdb.ornl.gov

- MSTDB-TP has undergone 2 major expansion efforts:**

- 1.0 to 2.0 (68 entries to 273 entries)
- 2.0 to 2.1 (273 entries to 311 entries)

- These expansions include replacements of old data with new data:**

- E.g. recent literature and relevant mixture expansion coefficients understood

- MSTDB-TP is being expanded with new releases**

- This includes new publications and higher order system data that exist in literature and need evaluated
- MSTDB-TP will also include new data of new systems as it is published

- MSTDB-TP is intending on including surface tension data in the future**

- There is a significant body of literature already evaluated and tabulated

Pure:

Salt	Measurements			
	ρ	μ	κ	c_p
AlCl ₃	1	1	0	1
BeCl ₂	1	0	0	0
BeF ₂	1	1	1	1
CaCl ₂	1	1	1	1
CaF ₂	1	1	1	1

Binary:



Ternary:

Salt	Measurements			
	ρ	μ	κ	c_p
KCl-LiCl-NaCl	4	0	0	0
LiCl-NaCl-AlCl ₃	10	10	0	0
LiF-BeF ₂ -ThF ₄	3	2	0	0
LiF-BeF ₂ -ZrF ₄	1	0	0	0
LiF-NaF-BeF ₂	1	1	0	0
LiF-NaF-KF	1	1	1	1
LiF-BeF ₂ -UF ₄	36	36	0	0
NaF-BeF ₂ -UF ₄	79	71	0	0
NaF-KF-BeF ₂	1	1	0	0
NaF-KF-MgCl ₂	1	0	0	0
NaF-KF-UF ₄	1	1	1	1
NaF-KF-ZrF ₄	1	1	0	0
NaF-LiF-BeF ₂	4	4	0	0
NaF-LiF-ZrF ₄	10	1	0	1
NaF-ZrF ₄ -UF ₄	5	3	2	3
RbF-ZrF ₄ -UF ₄	2	2	1	1

The MSR Campaign is funding the application of quality rankings to data in MSTDB-TP as well as duplicate data

- In collaboration with ANL – Melissa Rose + team
- Increased confidence in recommended data sets
- Better characterization of uncertainty
- Improved visibility about data selection process

NpF ₃	0	0	0	0
PuCl ₃	0	0	0	1
PuF ₃	0	0	0	1
SrCl ₂	1	1	1	0
SrF ₂	1	1	1	0
ThCl ₄	1	0	0	0
ThF ₄	1	0	0	0
UCl ₃	1	0	0	1
UCl ₄	1	0	0	0
UF ₃	0	0	0	1
UF ₄	1	1	0	1
ZrCl ₄	1	1	0	0
ZrF ₄	1	0	0	0



Quaternary:

Salt	Measurements			
	ρ	μ	κ	c_p
LiF-BeF ₂ -UF ₄ -ThF ₄	1	1	0	0
LiF-BeF ₂ -ZrF ₄ -UF ₄	1	0	0	0
NaF-LiF-BeF ₂ -UF ₄	1	1	0	0
NaF-LiF-KF-UF ₄	2	2	1	1
NaF-LiF-ZrF ₄ -UF ₄	1	1	0	1

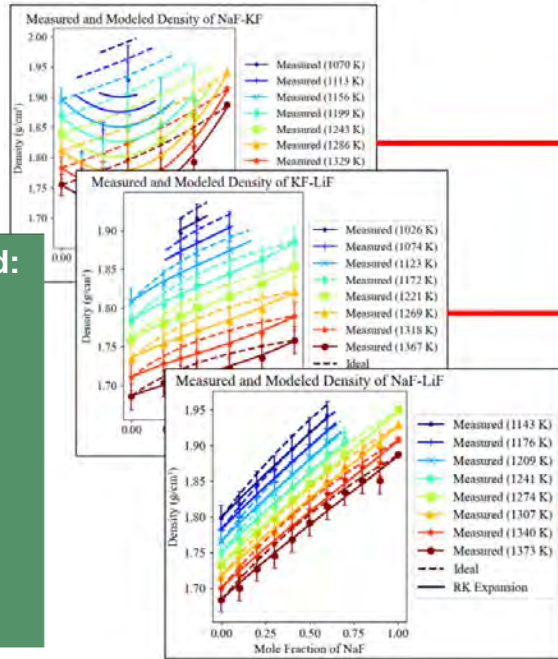
Redlich-Kister Density Models

RK Parameters Generated:

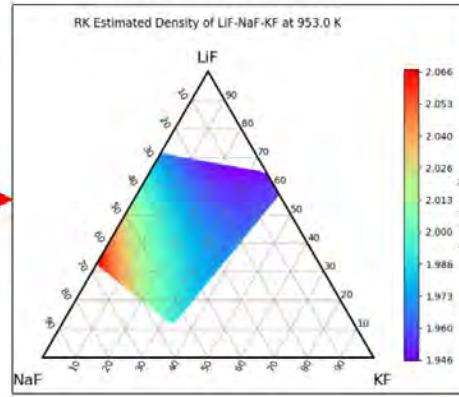
Fluorides
 LiF-BeF₂, LiF-KF, NaF-KF,
 NaF-LiF, NaF-ThF₄, LiF-
 UF₄, LiF-ZrF₄, LiF-ThF₄,
 NaF-ZrF₄, NaF-UF₄

Chlorides
 LiCl-KCl, KCl-MgCl₂, KCl-
 NaCl, KCl-UCl₃, KCl-UCl₄,
 NaCl-LiCl

Pseudobinary Interaction Determination



Higher Order System Model via Muggianu Interpolation



- **Motivation: We can only make measurements across the national laboratories so fast**

- Limited time and funding
- There are countless possible pseudo-ternary+ systems which may be of interest
- Therefore, implementing Redlich-Kister framework with Muggianu extrapolation need estimation techniques to predict thermophysical properties of higher-order systems

Chemical Engineering Science 247 (2021) 117906

Contents lists available at ScienceDirect

Chemical Engineering Science

Journal homepage: www.elsevier.com/locate/ces

Empirical estimation of densities in NaCl-KCl-UCl₃ and NaCl-KCl-YCl₃ molten salts using Redlich-Kister expansion

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^aMaterials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, TN, USA

HIGHLIGHTS

- Density of NaCl-KCl, NaCl-YCl₃, KCl-YCl₃, NaCl-UCl₃, and KCl-UCl₃ molten salts were modeled.
- Temperature and composition dependent Redlich-Kister expansion is used in the model.
- NaCl-KCl-UCl₃ and NaCl-KCl-YCl₃ molten salt densities were successfully estimated.
- Muggianu interpolation scheme was used to estimate the ternary molten salt densities.

GRAPHICAL ABSTRACT

Chemical Engineering Science 369 (2022) 117954

Contents lists available at ScienceDirect

Chemical Engineering Science

Journal homepage: www.elsevier.com/locate/ces

Application of the Redlich-Kister expansion for estimating the density of molten fluoride pseudo-ternary salt systems of nuclear industry interest²¹

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HIGHLIGHTS

- Several molten fluoride pseudo-ternary salt system densities were estimated.
- The estimation is based on Redlich-Kister expansion and Muggianu interpolation.
- This method generally outperforms estimation by additive molar volumes.
- Some unmeasured pseudo-binary system densities were also estimated.

More Information Available in FY23 Status Milestone Report⁷ – Available this summer

Ongoing Effort: Viscosity Estimation

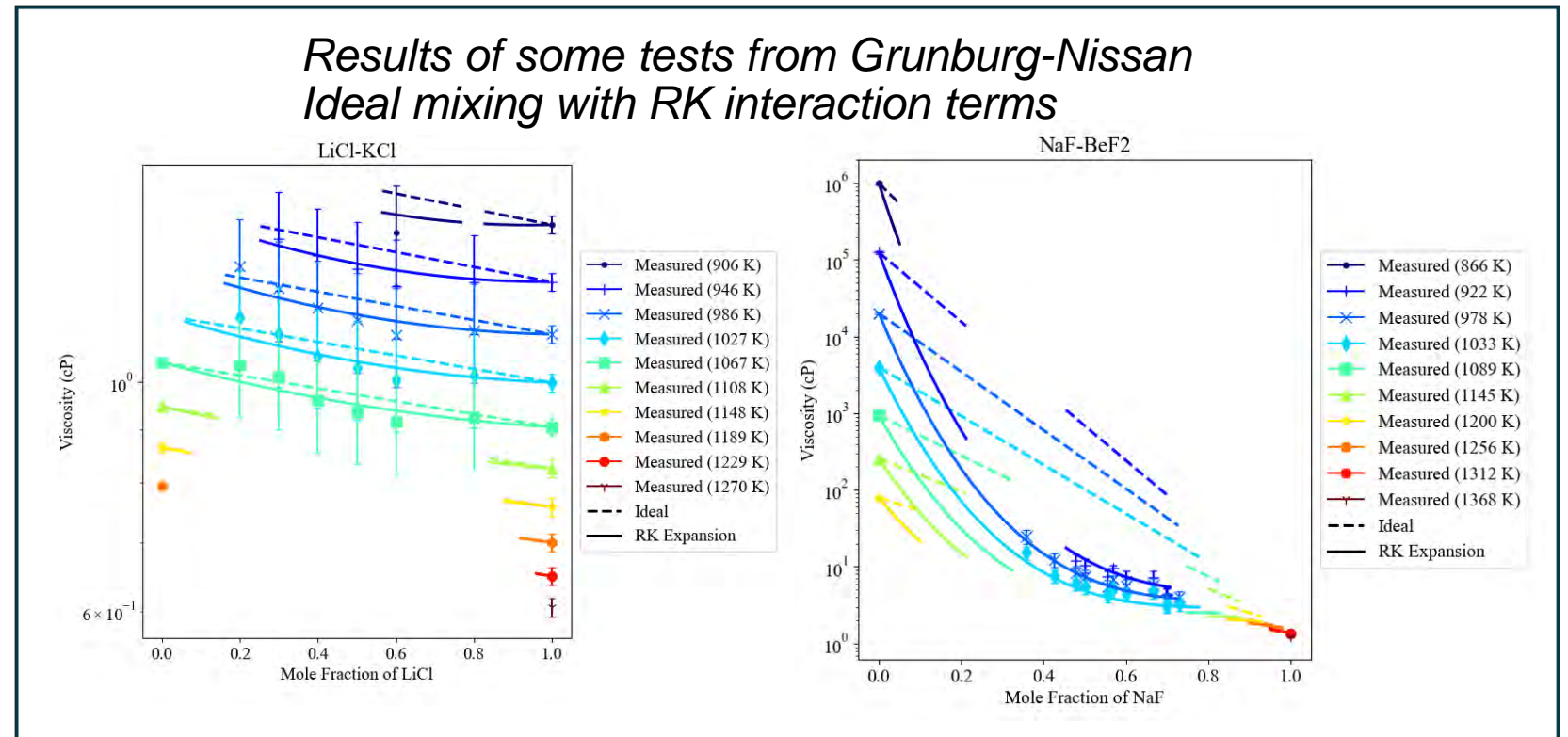
- **Viscosity Estimation is proving to be more challenging than density**

- Multiple manners by which one may model ideal or non-ideal terms

- Ideal mixing: Gambill method, Grunburg-Nissan rule, Katti-Chaudhri rule
- Non-Ideal terms: RK, modified Grunburg-Nissan

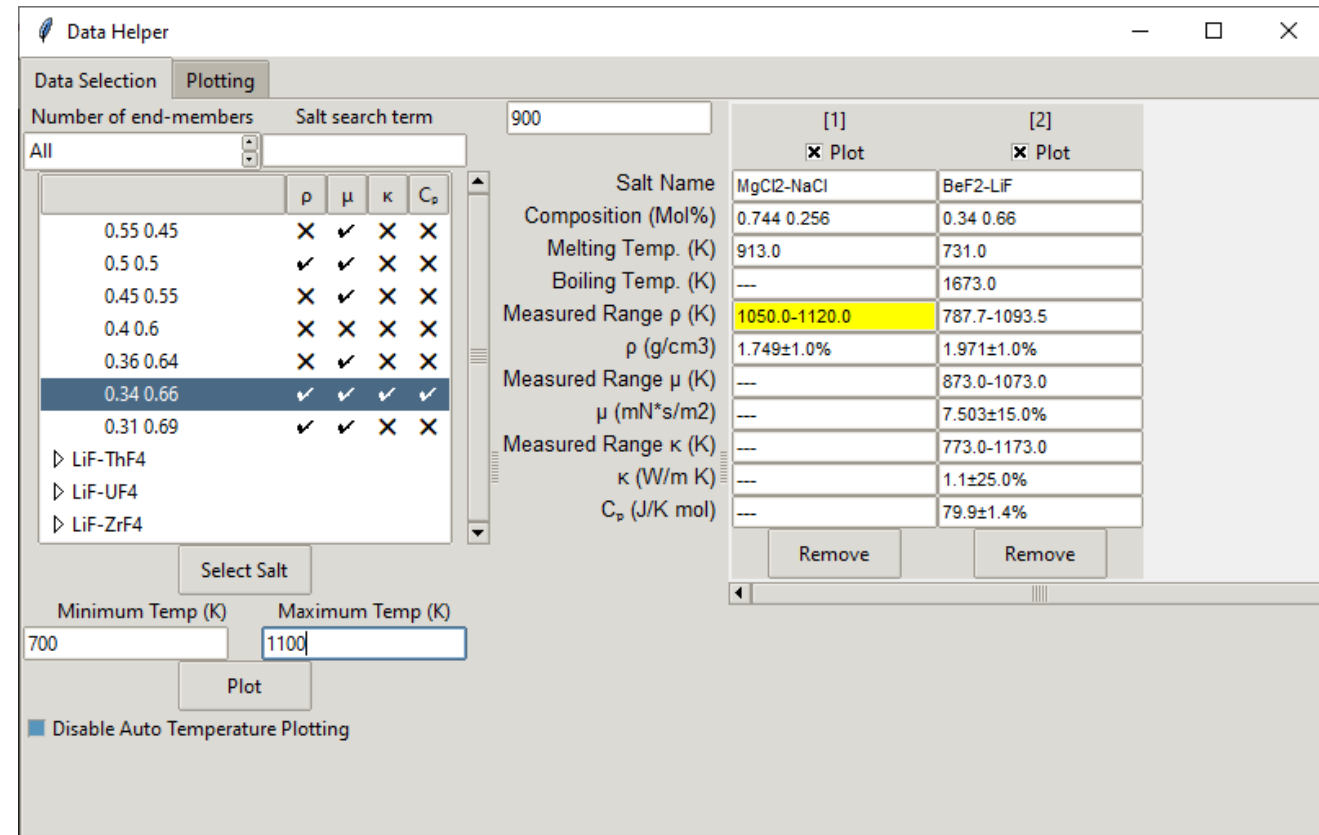
- We are investigating ALL possible methods and different orders of fit to find a consistent, generalizable method

- The ultimate goal is to validate extrapolative capabilities with higher-order systems



User/Modeler Accessibility through GUI or API

- **MSTDB-TP is a large, difficult to navigate .csv file**
- **Saline is the MSTDB-TP API**
 - Provides a stable C++ interface for obtaining supported properties (density, viscosity, heat capacity, and thermal conductivity)
 - Designed for integration with other NEAMS codes
- **The MSTDB-TP GUI allows for easier navigation and visualization of the data**



The screenshot shows the 'Data Helper' application window. It has two tabs: 'Data Selection' and 'Plotting'. The 'Data Selection' tab is active, showing a table of salts with columns for composition (p, μ, κ, C_p) and checkboxes for each property. The salt '0.34 0.66' is selected. Below the table are input fields for 'Minimum Temp (K)' (700) and 'Maximum Temp (K)' (1100), and a 'Plot' button. The 'Plotting' tab is also visible, showing a comparison of two salts: MgCl2-NaCl and BeF2-LiF. The 'Measured Range ρ (K)' for MgCl2-NaCl is highlighted in yellow.

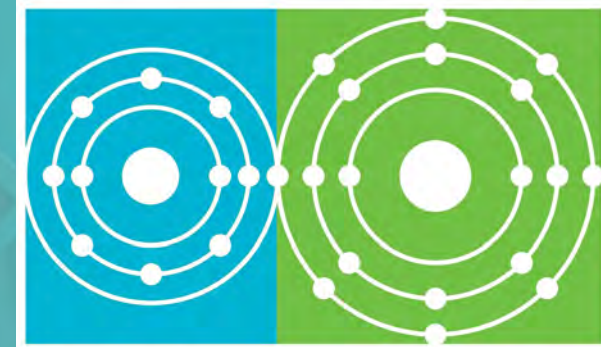
	[1]	[2]
	✗ Plot	✗ Plot
Salt Name	MgCl2-NaCl	BeF2-LiF
Composition (Mol%)	0.744 0.256	0.34 0.66
Melting Temp. (K)	913.0	731.0
Boiling Temp. (K)	---	1673.0
Measured Range ρ (K)	1050.0-1120.0	787.7-1093.5
ρ (g/cm3)	1.749±1.0%	1.971±1.0%
Measured Range μ (K)	---	873.0-1073.0
μ (mN*s/m2)	---	7.503±15.0%
Measured Range κ (K)	---	773.0-1173.0
κ (W/m K)	---	1.1±25.0%
C _p (J/K mol)	---	79.9±1.4%

Contact: Nick Termini termininc@ornl.gov

Acknowledgements

NEAMS

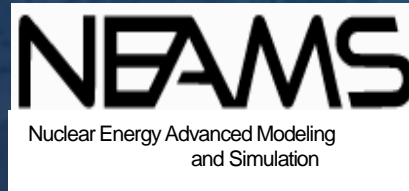
This work is directly funded by the Nuclear Energy Advanced Modeling and Simulation (NEAMS) Program and the Advanced Reactors Program under the Office of Nuclear Energy. The authors would like to thank Ted Besmann for guidance on MSTDB-TP development decisions.



Molten Salt Reactor
P R O G R A M

Thank you

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