

U.S. DEPARTMENT OF
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Thermochemical and Thermophysical Property Development for the Molten Salt Databases at LANL

Marisa Monreal

Inorganic, Isotope, and Actinide Chemistry – Chemistry Division

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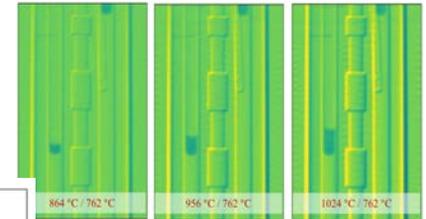
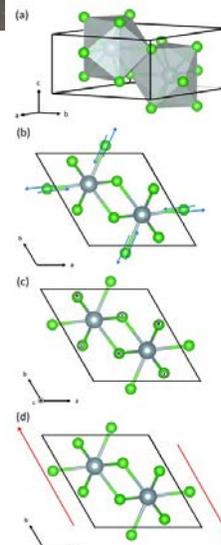
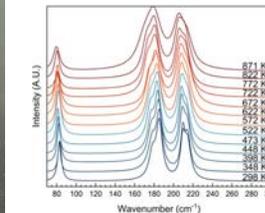
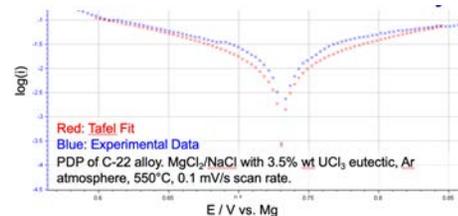


Annual MSR Campaign Review Meeting 16-18 April 2024

LA-UR-24-23405

LANL Actinide-Molten Salt Chemistry and Properties Research

- **Salt systems:**
 - Actinides: uranium, thorium, plutonium
 - Chlorides; expanding to fluorides, beryllium
- **Research:**
 - Chemistry & thermophysical properties
 - Experiment and modeling
 - Studies across length scales
- **Benefitting communities:**
 - Pyroprocessing, nuclear energy, fundamental actinide science, global security/nonproliferation



LANL Actinide-Molten Salt Chemistry and Properties Research

• Researchers:

- Graduate students, post-docs, scientists, and engineers
- Chemistry, materials science, earth and environmental science, theoretical

• Collaborations:

- National Laboratories, universities, industry, NEAMS, SciDAC, FUTURE EFRC

• Sponsors/Projects:

- MSR Campaign, LDRD, GAIN, Technology Commercialization Fund (TCF), IRP, NEUP

Scientists & Engineers

Matt Jackson
Scott Parker
David Andersson
Alex Long
Hongwu Xu
Sven Vogel
Karla Erickson
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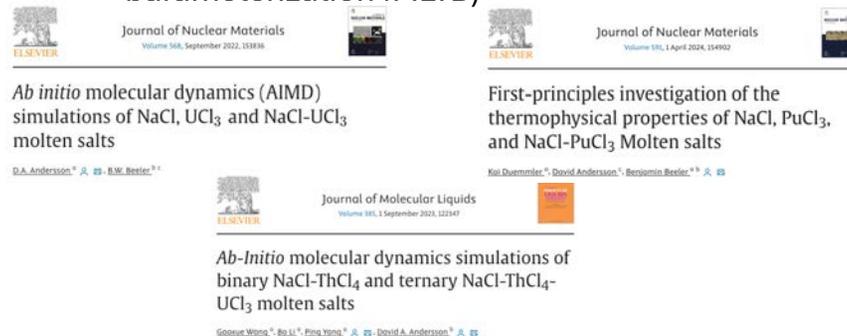


Adam Altenhof

LANL Actinide-Molten Salt Chemistry and Properties Research

Properties	Experimental Techniques
Density	Neutron Radiography, Conventional (Push-rod) Dilatometry
Viscosity	Dynamic Neutron Radiography, Rotational Viscometry
Melting Point/Phase Diagram, Heat Capacity	Differential Scanning Calorimetry (DSC)
Corrosion	Electrochemistry, Exposure Tests
Heat of Dissolution, Enthalpy of Mixing, Heat Capacity	Drop Calorimetry
Local Structure	Pair Distribution Function (PDF) Analysis, Raman Spectroscopy, Electrochemistry
Synthesis & Characterization	Inorganic halide synthesis, SEM, Melting Point (DSC), pXRD, SS-NMR Spectroscopy

- **Experimental efforts & strategy:**
 - Method development, uncertainty reduction
 - Lower and higher throughput methods
 - Non-rad → U , Th → Pu
 - Iterate with mod-sim
- **Modeling and Simulation:**
 - David Andersson, Ping Yang, Gaoxue Wang
 - Ab initio molecular dynamics (AIMD) simulations; density functional tight binding (DFTB) modeling, machine-learning-augmented DFTB parameterization (MLTB)



LANL MSR Campaign Work

1. PuCl₃-NaCl salts (FY22-present)

- Eutectic (36% PuCl₃ 64% NaCl)
- Ternary: PuCl₃ in UCl₃-NaCl eutectic (<10% PuCl₃)
 - **Density:** LANSCE neutron radiography method
 - **Heat capacity (C_p):** Drop calorimetry for integral C_p

2. Uranium-containing salts (FY24)

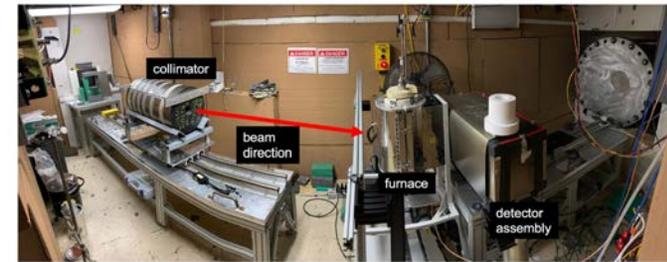
- **Synthesis & characterization of uranium halides**
 - UCl₃, UF₄; UF₃, UI₃
 - LiF-NaF-KF-UF₄: ΔH_{fus}
 - UF₄: C_p >350K to 1173K
 - NaCl-MgCl₂-UCl₃: phase equilibria
- Ted Besmann-recommended / MSTDB*

LANL MSR Campaign Work: PuCl₃-NaCl Salts

Density: LANSCE neutron radiography



- ✓ **Inter-lab collaboration:** INL and LANL measured the same material: INL-synthesized PuCl₃-NaCl eutectic
 - INL – Toni Karlsson: Shipped material to LANL: 10g shipped and introduced into glovebox line in the Plutonium Facility at LANL (PF-4)
- ✓ Upon receipt, material purity checked by melting point (DSC)



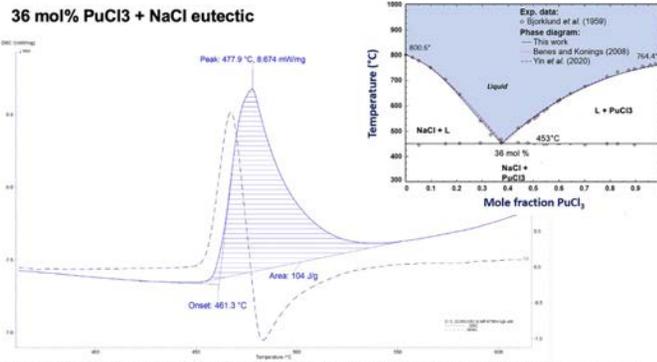
Flight Path 5 at Los Alamos Neutron Science Center (LANSCE)



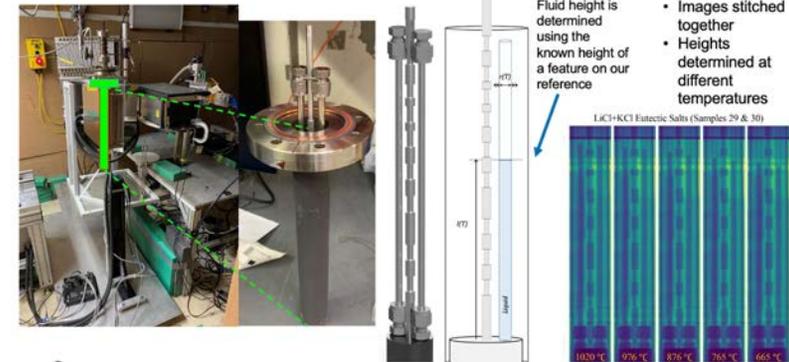
PuCl₃ + NaCl eutectic salt (INL)



Sample Stage

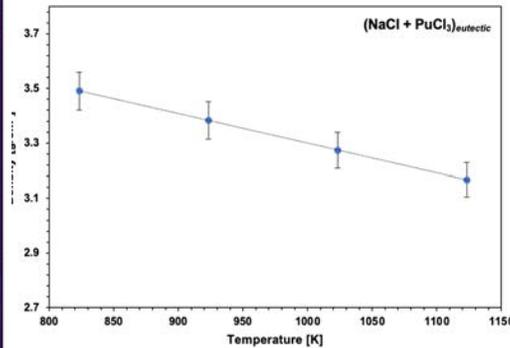
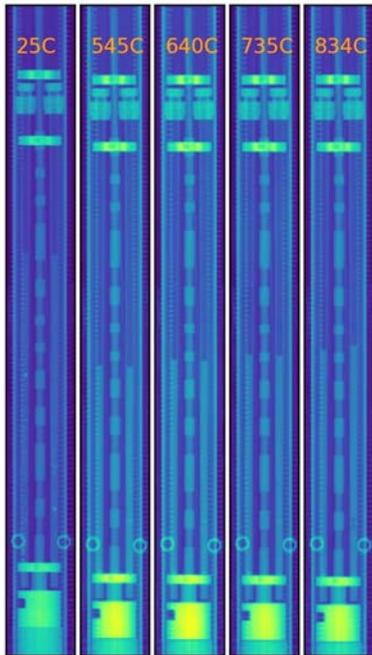


This work: $T_{\text{melt, onset}} = 461 \pm 2 \text{ } ^\circ\text{C}$, $T_{\text{melt, peak}} = 478 \pm 2 \text{ } ^\circ\text{C}$, $\Delta H_{\text{fusion}} = 104 \text{ J/g}$

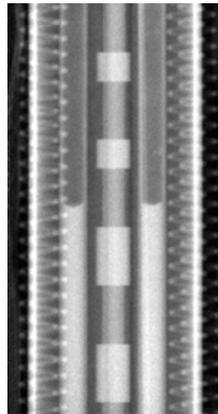


LANL MSR Campaign Work: $\text{PuCl}_3\text{-NaCl}$ Salts

Density: LANSCE neutron radiography



Temperature [°C]	Density [g/cm ³]
550	3.49
650	3.38
750	3.27
850	3.17



Alex Long, Scott Parker,
Brendt Wohlberg

- ✓ Density by neutron radiography measurements completed on a set of $\text{PuCl}_3\text{-NaCl}$ and $\text{PuCl}_3\text{-UCl}_3\text{-NaCl}$ samples; preliminary image analysis and density values generated
- **Current work: Ensuring accuracy of volume measurement from images** → Accounting for filled and void spaces
 1. Modeler working on volume estimation, including modeling bubbles/voids
 - Potential future applications: analysis of volatiles in salt, surface tension
 2. Analyzing change in relative contrast
 - Average column intensity, compare to full intensity, any differences thus equal voids
- **Current work: Next steps for the LANSCE-measured samples**
 1. Post-mortem characterization (XRD)
 2. If pure, re-constituting into new compositions, making new measurements (more density, drop cal)

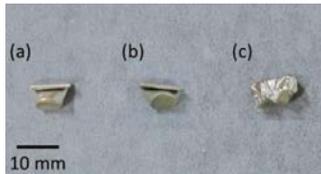
LANL MSR Campaign Work: PuCl₃-NaCl Salts

Integral Heat Capacity (C_p): Drop Calorimetry

- **En route to integral C_p:** with LDRD funding, a new methodology for measuring the ΔH_{mix} of molten salt systems was developed using Setaram AlexSYS-800 calorimeter.¹

Advantages/improvements:

- Elimination of problems from material-atmosphere interactions
- Minimal salt-crucible interactions
- No further mixing needed after being introduced into the calorimeter

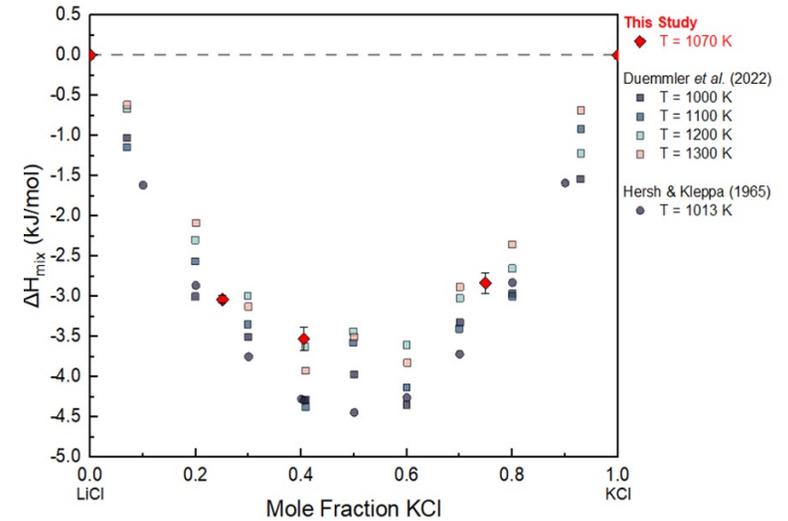


LiCl-KCl eutectic sample:
 (a) before the calorimetric measurement
 (b) after the measurement
 (c) after sectioning

- **Recent results: PuCl₃-NaCl eutectic integral C_p**
 - Used INL-synthesized material
 - **Good match with INL-measured C_p (by DSC); publication forthcoming**



Andrew Strzelecki (former LANL PD), Hongwu Xu



Sample	Temperature Room (K)	Temperature Calorimeter (K)	Experimental ΔH_{mix} (kJ/mol)
0.251KCl – 0.749LiCl	297.48 ± 0.11	1070.39 ± 0.01	-3.04 ± 0.05
KCl-LiCl eutectic	297.73 ± 0.15	1070.51 ± 0.00	-3.52 ± 0.15
0.749KCl – 0.251 LiCl	297.62 ± 0.27	1070.41 ± 0.01	-2.83 ± 0.13

(1) Strzelecki, A., Cockreham, C., Parker, S., Mann, S., Lhermitte, C., Wu, D., Guo, X., Monreal, M. J., Jackson, J. M., Mitchell, J., Boukhalifa, H., Xu, H. "A New Methodology for Measuring the Enthalpies of Mixing of Molten Salts Using High Temperature Drop Calorimetry", Rev. Sci. Inst., 2024, 95, 014103.

LANL MSR Campaign Work: Uranium-containing salts

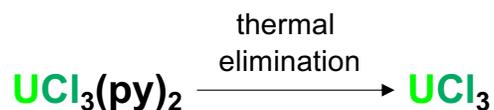
Synthesis of Uranium Halides (chlorides, fluorides, iodide)

Current work: Synthesis of *pure, isolable* actinide chlorides and fluorides to enable property research

- Both conventional & novel synthetic routes
- Characterization techniques to confirm purity (e.g., SS-NMR; pXRD; DSC)
 - Impurities: other actinide species; water; products of rxn with water

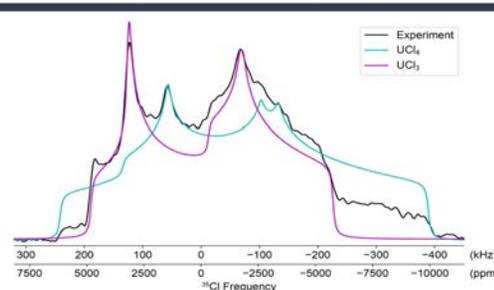


With LDRD funding, novel route to UCl_3 developed²:

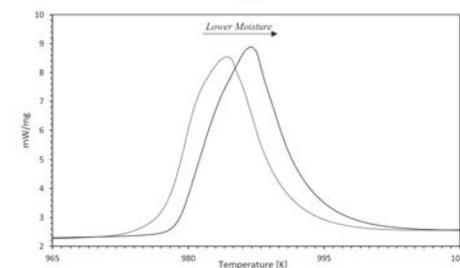
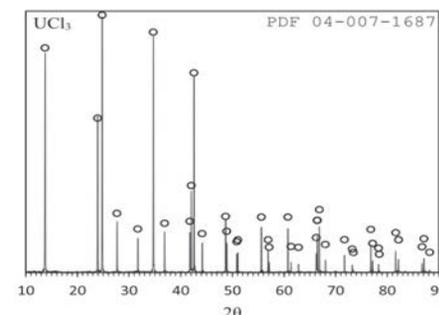


MSR Campaign work: translate novel route learnings to other halides; execute conventional routes

258 (Ground, 66 : 33 UCl_3 : UCl_4)



SS-NMR: Adam Altenhof (PD),
Harris Mason



XRD, melting point: Scott
Parker (and more!)

(2) Erickson, K., Parker, S., Monreal, M. J. "Thermal Elimination of Pyridine from a Uranium Trichloride Precursor", *Chem. Methods*, **2024**, e202300052.

LANL MSR Campaign Work: Uranium-containing salts

Synthesis of Uranium Halides (chlorides, fluorides, iodide)

Current work: New synthesis of UF_4 under development



UF_4

Scale: ≤ 1.5 g UF_4

Workup 1: wash

Dry in desiccator

2% organic
1% iodine

Heat at 150 °C (hours)

1-2% organic
<1% iodine

Workup 2: no wash

Heat at 150 °C

1-2% organic
<1% iodine

Heat at 250 °C

1% organic
0% iodine

*Small amounts of
organic and iodine
byproduct*

*Nicolas Capra (PD),
Karla Erickson*

*+ Incoming Seaborg 2024 Summer
Student*

*Scott Parker, Hannah Patenaude
New addition: Kristen Pace*

Current work:

- Completing workup refinement, scale up
- Collecting characterization data
 - ^{19}F Solid-State NMR Spectroscopy (first for UF_4 !)
 - Others?
- Manuscript preparation: To be submitted by end of CY24
- Exploring other novel fluorination routes
- Next steps:
 - Synthetic efforts: UF_3 , UI_3
 - Handoff to properties measurements
 - $LiF-NaF-KF-UF_4$: ΔH_{fus}
 - UF_4 : Cp >350K to 1173K
 - $NaCl-MgCl_2-UCl_3$: phase equilibria

Looking Forward: Collaborations and Building the Pipeline

Internal Collaborations:

- **LANL-led SciDAC**
 - Scientific Discovery through Advanced Computing program to advance modeling behavior and properties of structural materials under molten salt conditions
- **Nuclear Energy Advanced Modeling and Simulation (NEAMS)****
- **LANL Civilian Nuclear Program Director: Chris Stanek****
- **LANL-led EFRC: FUTURE**
 - Studying corrosion under irradiation

Upcoming 2024 presentations:

- MRS Spring Meeting
- Fall ACS
- Actinide Separations
- Plutonium Futures
- MSR Workshop

LANL-University Collaborations:

- **University of Utah** (Simpson)
 - Electrochemistry, vapor pressure, student pipeline
- **UC Berkeley** (Scarlat, Fratoni)
 - Molten salt round robin, fluoride & beryllium salt expertise, student pipeline, **IRP**
- **MIT** (Khaykovich)
 - Pair distribution function analysis, new **NEUP just announced!**
- **Oregon State University, Texas A&M, UNLV**
 - Student pipeline—viscosity, materials corrosion, electrochemistry

Technical Work Scope Identifier No. IRP-NEAMS-1

Bridging the gap between experiments and modeling to improve the design of molten salt reactors

Massimiliano Fratoni, Mark Asta, Peter Hosemann, Raluca Scarlat—University of California, Berkeley

Izabela Szlufarska—University of Wisconsin-Madison

Alexandra Navrotsky, Hongwu Xu—Arizona State University

Marisa Monreal—Los Alamos National Laboratory

Abdalla Abou Jaoude—Idaho National Laboratory

Carolyn Burns, Thomas Hartmann—Pacific Northwest National Lab

Nader Satvat—Kairos Power, LLC

Karl Britsch—TerraPower, LLC



Redox potential, ionic speciation, and separation and recovery challenges from molten salts containing actinides and fission products

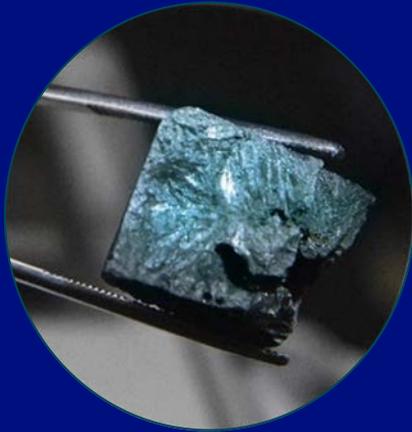
PI: Boris Khaykovich (MIT)

Collaborators: Stephen Lam (University of Massachusetts), David Sprouster (Stony Brook University), Anatoly Frenkel (Brookhaven National Laboratory), Marisa Monreal (Los Alamos National Laboratory), and Joanna McFarlane (Oak Ridge National Laboratory)

Program: Nuclear Fuel Recycle Technologies

New Plutonium R&D Capability: **Plutonium Science Laboratory (“PluS Lab”)**

-- Gram-scale, non-irradiated materials --



PLUTONIUM CAPABILITY 1:
Molecular chemistry &
materials science

ENVIRONMENT 1:
O₂- and H₂O-free

PLUTONIUM CAPABILITY 2:
Aqueous chemistry

ENVIRONMENT 2:
Water solutions

PLUTONIUM CAPABILITY 3:
Molten salt science

ENVIRONMENT 3:
High-temperature (400 °C-
900°C), O₂- and H₂O-free

Leadership Team: Marisa Monreal and Karla Erickson
Sponsor: Nonproliferation Stewardship Program (NSP)

Acknowledgements

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Adam Altenhof
Dylan Tharpe Eralie

University Collaborators, Visiting Students

University of Utah, MIT, UC Berkeley, OSU,
UNLV, Texas Tech

Funding:

Molten Salt Reactor
Campaign

LANL Laboratory Directed
Research and Development
(LDRD) Directed Research
Project #20210113DR

Gateway for Accelerated
Innovation in Nuclear (GAIN)
#NE-21-25117 with
TerraPower

Technology
Commercialization Fund
(TCF) with Kairos Power



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Thank you

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